Deleted: 3/6/2008

=> fil reg
FILE 'REGISTRY' ENTERED AT 11:56:55 ON 06 MAR 2008
USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.
PLEASE SEE "HELP USAGETERMS" FOR DETAILS.
COPYRIGHT (C) 2008 American Chemical Society (ACS)

Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 4 MAR 2008 HIGHEST RN 1006657-22-2 DICTIONARY FILE UPDATES: 4 MAR 2008 HIGHEST RN 1006657-22-2

New CAS Information Use Policies, enter HELP USAGETERMS for details.

TSCA INFORMATION NOW CURRENT THROUGH January 9, 2008.

Please note that search-term pricing does apply when conducting SmartSELECT searches.

REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

http://www.cas.org/support/stngen/stndoc/properties.html

=> d que stat 15 L3 S'

> 3 G1 Ak~0 4

0 G1 CH3 @11

VAR G1=H/11
NODE ATTRIBUTES:
CONNECT IS E2 RC AT 4
DEFAULT MLEVEL IS ATOM
GGCAT IS SAT AT 4
DEFAULT ECLEVEL IS LIMITED

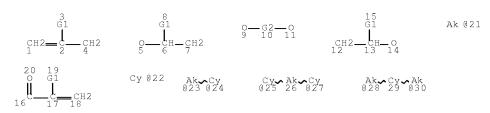
GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 11

STEREO ATTRIBUTES: NONE L5 62762 SEA FILE=REGISTRY SSS FUL L3

100.0% PROCESSED 394644 ITERATIONS SEARCH TIME: 00.00.04

62762 ANSWERS

=> d que stat 110 L10 STE

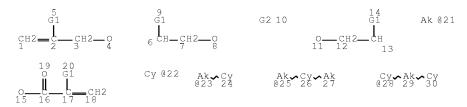


VAR G1=H/CH3
VAR G2=21/22/23-9 24-11/23-11 24-9/25-9 27-11/28-9 30-11
NODE ATTRIBUTES:
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 30

STEREO ATTRIBUTES: NONE

=> d que stat 115



VAR G1=H/CH3 VAR G2=21/22/23/25/28 NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 30

STEREO ATTRIBUTES: NONE

=> d que stat 123 L23 STF

3 Deleted: 3/6/2008

VAR G1=H/CH3 NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 14

STEREO ATTRIBUTES: NONE

=> d que stat 128







CH3 @11

VAR G1=H/11 NODE ATTRIBUTES: CONNECT IS E2 RC AT 4 DEFAULT MLEVEL IS ATOM GGCAT IS SAT AT 4 DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 11

STEREO ATTRIBUTES: NONE

L5 62762 SEA FILE=REGISTRY SSS FUL L3

L12 SCR 2026 OR 1313

L24







VAR G1=H/CH3 NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 14

STEREO ATTRIBUTES: NONE

200 SEA FILE=REGISTRY SUB=L5 SSS FUL L24 NOT L12

100.0% PROCESSED 945 ITERATIONS

SEARCH TIME: 00.00.01

200 ANSWERS

=> fil hcap FILE 'HCAPLUS' ENTERED AT 11:57:16 ON 06 MAR 2008 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2008 AMERICAN CHEMICAL SOCIETY (ACS)

Copyright of the articles to which records in this database refer is held by the publishers listed in the PUBLISHER (PB) field (available for records published or updated in Chemical Abstracts after December 26, 1996), unless otherwise indicated in the original publications. The CA Lexicon is the copyrighted intellectual property of the the American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited.

FILE COVERS 1907 - 6 Mar 2008 VOL 148 ISS 10 FILE LAST UPDATED: 5 Mar 2008 (20080305/ED)

New CAS Information Use Policies, enter HELP USAGETERMS for details.

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d his

(FILE 'HOME' ENTERED AT 09:49:59 ON 06 MAR 2008)

FILE 'HCAPLUS' ENTERED AT 09:50:09 ON 06 MAR 2008 E US20070135560/PN

L1 1 S E3

SEL RN

FILE 'REGISTRY' ENTERED AT 09:51:36 ON 06 MAR 2008 L2 $$5\ {\rm S}\ {\rm E1}{-}5$$

FILE 'LREGISTRY' ENTERED AT 10:08:44 ON 06 MAR 2008

FILE 'REGISTRY' ENTERED AT 10:21:15 ON 06 MAR 2008

L4 50 S L3

L5 62762 S L3 FUL

SAV L5 TEMP RED210/A

L6 1 S L2 AND L5

FILE 'LREGISTRY' ENTERED AT 10:37:56 ON 06 MAR 2008

L7 STR L8 STR

FILE 'REGISTRY' ENTERED AT 11:05:40 ON 06 MAR 2008

L9 50 S L7 SSS SAM SUB=L5

L10 STR L7

L11 46 S L10 SSS SAM SUB=L5

L12 SCR 2026 OR 1313

L13 3 S L10 NOT L12 SSS SAM SUB=L5

L14 55 S L10 NOT L12 SSS FUL SUB=L5

SAV L14 RED210S1/A

<u>3/11/2008</u> ,		10/588,210		
L16 2 S L L17 44 S L	L8 15 NOT L12 S 15 NOT L12 S L17 RED210S	SS FUL SUB=		
L19 0 S (L20 11097 S L	1637-38-1/RN L14 OR L17) 5 AND C2H4O 2 AND L20	AND C2H4O		
FILE 'LREGISTR L22 STR L23 STR L24 STR		T 11:27:31	ON 06 MAR 2008	
L26 158 S L SAV	23 NOT L12 S 23 NOT L12 S RED210S3/A	SS SAM SUB= SS FUL SUB= L26	L5 L5	
L28 200 S L	24 NOT L12 S 24 NOT L12 S L28 RED210S	SS FUL SUB=		
L30 39 S (L31 57 S (L26 OR L28) L14 OR L17) L26 OR L28) 29 OR L31	AND ETHOXY		
FILE 'HCAPLUS' L33	30	11:52:00 ON	06 MAR 2008	
FILE 'CAOLD' E L36	30	:52:36 ON 0	6 MAR 2008	
=> d 135 ibib abs h	itstr hitind	1-26		
L35 ANSWER 1 OF 26 ACCESSION NUMBER: DOCUMENT NUMBER: TITLE:	2007:9 147:28 Eugeno	35013 HCAP 5289 l-derived m	08 ACS on STN LUS <u>Full-text</u> onomers and acrylic p for biomedical uses	polymers
INVENTOR(S):	Rojo D	el Olmo, Lu	is; Vazquez Lasa, Mar	
PATENT ASSIGNEE(S):			rio, Julio; Deb, San de Investigaciones C	
SOURCE:	PCT In	t. Appl., 5 PIXXD2	0pp.	
DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. CO PATENT INFORMATION:	Patent Spanis			
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE

200702

5 Deleted: 3/6/2008

```
14
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA,
             CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,
             GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE,
             KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LV, LY,
             MA, MD, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM,
             PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SV,
             SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM,
             ZW
         RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU,
             IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR,
             BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD,
             TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM,
             ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
PRIORITY APPLN. INFO.:
                                            ES 2006-347
                                                                    200602
```

The eugenol-derivative monomers have vinyl and methacrylate groups and alkyl AB substituents, e.g., eugenyl methacrylate, ethoxy-eugenyl methacrylate and are prepared by reaction of eugenol or alkyl derivative with methacryloyl chloride. The eugenyl methacrylate or derivative polymers comprise the eugenyl methacrylate monomer or derivative and Me methacrylate or Et methacrylate and are prepared by radical polymerization using AIBN initiator, at 50-60°. Self-curable formulations comprise eugenyl methacrylate or derivative; Me methacrylate or Et methacrylate; an aromatic tertiary amine; a quinone inhibitor; acrylic polymer particles; 50-80% ZnO particles; benzoyl peroxide initiator; and x-ray contrast agents, selected from BaSO4, ZrO2, Ta oxide, Sr oxide, and organic compds. The self-curable composition is used by direct application and in-situ cure for temporary or permanent dental and bone reconstruction, i.e., vertebral bone setting, setting of osteoporotic fractures in minimally-invasive surgery.

312479-75-5P, Ethoxy-eugenyl methacrylate homopolymer RL: IMF (Industrial manufacture); PREP (Preparation) (eugenyl methacrylate monomers and copolymers and self-curable compns. for dental and bone reconstruction)

RN 912479-75-5 HCAPLUS

2-Propenoic acid, 2-methyl-, 2-[2-methoxy-4-(2-propen-1yl)phenoxy]ethyl ester, homopolymer (CA INDEX NAME)

CM 1

CRN 912479-73-3 CMF C16 H20 O4

912479-77-7P, Ethoxy-eugenyl methacrylate-ethyl methacrylate copolymer RL: IMF (Industrial manufacture); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses) (eugenyl methacrylate monomers and copolymers and self-curable compns. for dental and bone reconstruction) 912479-77-7 HCAPLUS RN

2-Propenoic acid, 2-methyl-, ethyl ester, polymer with 2-[2-methoxy-4-(2-propen-1-yl)phenoxy]ethyl 2-methyl-2-propenoate (CA INDEX NAME)

 $\mathbb{C}\mathbb{M}$ 1

912479-73-3 CRN C16 H20 O4

$$\begin{array}{c} {\rm H_{2}C} \\ {\rm Me} = \begin{array}{c} {\rm CH_{2}-$$

CM 2

CRN 97-63-2 CMF C6 H10 O2

912473-73-3P, Ethoxy-eugenyl methacrylate RL: IMF (Industrial manufacture); RCT (Reactant); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(monomer; eugenyl methacrylate monomers and copolymers and self-curable compns. for dental and bone reconstruction)

RN 912479-73-3 HCAPLUS CN

2-Propenoic acid, 2-methyl-, 2-[2-methoxy-4-(2-propen-1yl)phenoxy]ethyl ester (CA INDEX NAME)

$$\begin{array}{c} {\rm H_{2}C} \\ {\rm Me-C-C-O-CH_{2}-CH_{2}-O} \end{array}$$

63-7 (Pharmaceuticals)

Section cross-reference(s): 35

912479-74-4P, Eugenyl methacrylate homopolymer 912479-75-5P

, Ethoxy-eugenyl methacrylate homopolymer

RL: IMF (Industrial manufacture); PREP (Preparation)

(eugenyl methacrylate monomers and copolymers and self-curable compns. for dental and bone reconstruction)

912479-76-6P, Ethyl methacrylate-eugenyl methacrylate copolymer ΙT

912479-77-78, Ethoxy-eugenyl methacrylate-ethyl methacrylate copolymer

RL: IMF (Industrial manufacture); THU (Therapeutic use); BIOL

(Biological study); PREP (Preparation); USES (Uses)

3/11/200& (eugenyl methacrylate monomers and copolymers and self-curable

compns. for dental and bone reconstruction)

375856-97-6P, Eugenyl methacrylate 912473-73-3P,

4

Ethoxy-eugenyl methacrylate

RL: IMF (Industrial manufacture); RCT (Reactant); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); RACT (Reactant or reagent): USES (Uses)

(monomer; eugenyl methacrylate monomers and copolymers and self-curable compns. for dental and bone reconstruction)

REFERENCE COUNT:

AUTHOR(S):

THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN

THE RE FORMAT

L35 ANSWER 2 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:1079275 HCAPLUS Full-text

DOCUMENT NUMBER: 146:62955

Solution- and Solid-Phase, Modular Approaches TITLE: for Obtaining Different Natural Product-Like Polycyclic Architectures from an Aminoindoline

Scaffold for Combinatorial Chemistry Reddy, P. Thirupathi; Quevillon, S.; Gan, Zhonghong; Forbes, Nauzer; Leek, Donald M.;

Arya, Prabhat

CORPORATE SOURCE: Steacie Institute for Molecular Sciences,

National Research Council of Canada, Ottawa, ON,

K1A OR6, Can.

SOURCE: Journal of Combinatorial Chemistry (2006), 8(6),

856-871

CODEN: JCCHFF; ISSN: 1520-4766

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 146:62955

* STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT *

With the goal of developing a modular approach leading to different indoline alkaloid natural-product-like tricyclic derivs. having an unsatd. lactam, an aminoindoline-based bicyclic scaffold I was obtained from aminoindole II. The selective deprotection of the indoline N-Teoc or benzylic NHAlloc in compound I, followed by N-acryloylation and then subjection to a ring-closing metathesis reaction, successfully led to obtaining two different architectures having an unsatd. lactam functionality, e.g. III. This modular solution-phase methodol. was then developed on solid phase. To achieve this objective, the aminoindoline bicyclic scaffold having an addnl. hydroxyl group could be immobilized onto the solid support using alkylsilyl linker-based polystyrene macrobeads. By applying a ring-closing metathesis approach, a tricyclic derivative with seven-membered-ring unsatd. lactam and a tricyclic derivative with eight-membered-ring unsatd. lactam were then obtained from in a number of steps.

916658-23-6P 916658-38-3P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(solution and solid-phase preparation of natural product-like polycyclic compds. from aminoindole scaffold for combinatorial chemical)

916658-23-6 HCAPLUS RN

3/11/2008, 10/588,210 Deleted: 3/6/2008

Carbamic acid, N-[(2S, 3S)-2-[(1R)-1-(acetyloxy)-3-buten-1-y1]-2,3-(acetyloxy)-3-(acetdihydro-5-[(2-methoxyethoxy)methoxy]-1-(1-oxo-2-propen-1-yl)-1Hindol-3-yl]-, 2-propen-1-yl ester (CA INDEX NAME)

Absolute stereochemistry.

916658-38-3 HCAPLUS CN Carbamic acid, N-[(2S,3S)-2-[(1S)-1-(acetyloxy)-3-buten-1-y1]-2,3-(acetyloxy)-3- $\label{lem:dihydro-5-[(2-methoxyethoxy)methoxy]-1-(1-oxo-2-propen-1-yl)-1H-} \\$

indol-3-yl]-, 2-propen-1-yl ester (CA INDEX NAME)

Absolute stereochemistry.

```
31-5 (Alkaloids)
    850559-86-3P 916658-22-5P 916658-23-6P
                                               916658-26-9P
    916658-28-1DP, resin-bound
                                916658-28-1P
                                               916658-29-2DP,
                  916658-30-5DP, resin-bound
                                               916658-31-6DP,
    resin-bound
                  916658-32-7DP, resin-bound
                                               916658-35-0P
    resin-bound
    916658-37-2P 916658-33-3P 916658-39-4P
                                               916658-40-7P
     916658-41-8P
                  916658-42-9P 916658-43-0P 916658-44-1P
                  916658-46-3DP, resin-bound
    916658-45-2P
                                               916658-47-4DP,
    resin-bound 916658-48-5DP, resin-bound
                                               916658-49-6DP,
                 916658-50-9DP, resin-bound
                                               916658-51-0DP,
    resin-bound
    resin-bound
                  916658-52-1DP, resin-bound
    RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);
    RACT (Reactant or reagent)
        (solution and solid-phase preparation of natural product-like polycyclic
        compds. from aminoindole scaffold for combinatorial chemical)
REFERENCE COUNT:
                              THERE ARE 28 CITED REFERENCES AVAILABLE
                        28
                              FOR THIS RECORD. ALL CITATIONS AVAILABLE
                              IN THE RE FORMAT
```

L35 ANSWER 3 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:935643 HCAPLUS Full-text DOCUMENT NUMBER: 145:425614 From Natural Products to Polymeric Derivatives TITLE:

of "Eugenol": A New Approach for Preparation of

Dental Composites and Orthopedic Bone Cements
AUTHOR(S): Rojo, Luis; Vazquez, Blanca; Parra, Juan; Lopez

Bravo, Antonio; Deb, Sanjukta; San Roman, Julio Institute of Polymer Science and Technology,

CSIC, Madrid, 28006, Spain

SOURCE: Biomacromolecules (2006), 7(10), 2751-2761

CODEN: BOMAF6; ISSN: 1525-7797

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

CORPORATE SOURCE:

Polymers with eugenol moieties covalently bonded to the macromol. chains were synthesized for potential application in orthopedic and dental cements. First, eugenol was functionalized with polymerizable groups. The synthetic methods employed afforded two different methacrylic derivs., where the acrylic and eugenol moieties were either directly bonded, eugenyl methacrylate (EgMA), or separated through an oxyethylene group, ethoxyeugenyl methacrylate (EEgMA). A typical Fisher esterification reaction was used for the synthesis of EgMA and EEqMA, affording the desired monomers in 80% yields. Polymerization of each of the novel monomers, at low conversion, provided soluble polymers consisting of hydrocarbon macromols. with pendant eugenol moieties. At high conversions only cross-linked polymers were obtained, attributed to participation of the allylic double bonds in the polymerization reaction. In addition, copolymers of each eugenol derivative with Et methacrylate (EMA) were prepared at low conversion, with the copolymn. reaction studied by assuming the terminal model and the reactivity ratios determined according to linear and nonlinear methods. The values obtained were rEgMA = 1.48, rEMA = 0.55 and rEEgMA = 1.22, rEMA = 0.42. High mol. weight polymers and copolymers were obtained at low conversion. Anal. of thermal properties revealed a Tg of 95 °C for PEgMA and of 20 °C for PEEgMA and an increase in the thermal stability for the eugenol derivs. polymers and copolymers with respect to that of PEMA. Water sorption of the copolymers was found to decrease with the eugenol derivative content. Both monomers EqMA and EEqMA showed antibacterial activity against Streptococcus mutans, producing inhibition halos of 7 and 21 mm, resp. Finally, cell culture studies revealed that the copolymers did not leach any toxic eluants and showed good cellular proliferation with respect to PEMA. This study thus indicates that the eugenyl methacrylate derivs. are potentially good candidates for dental and orthopedic cements. ΙT 912479-75-5P 912479-77-7P

RL: BSU (Biological study, unclassified); PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(preparation of dental composites and orthopedic bone cements from polymeric derivs. of the natural product eugenol)

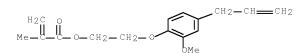
912479-75-5 HCAPLUS

2-Propenoic acid, 2-methyl-, 2-[2-methoxy-4-(2-propen-1-yl)phenoxy]ethyl ester, homopolymer (CA INDEX NAME)

CM 1

RN CN

> CRN 912479-73-3 CMF C16 H20 O4



RN 912479-77-7 HCAPLUS

2-Propenoic acid, 2-methyl-, ethyl ester, polymer with 2-[2-methoxy-4-(2-propen-1-yl)phenoxy]ethyl 2-methyl-2-propenoate (CA INDEX NAME)

 $\mathbb{C}\mathbb{M}$ 1

CRN 912479-73-3 CMF C16 H20 O4

2 ${\tt CM}$

CRN 97-63-2 CMF C6 H10 O2

912479-73-32 ΙT

> RL: BSU (Biological study, unclassified); RCT (Reactant); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); RACT (Reactant or reagent) (preparation of dental composites and orthopedic bone cements from polymeric derivs. of the natural product eugenol) 912479-73-3 HCAPLUS

RN CN2-Propenoic acid, 2-methyl-, 2-[2-methoxy-4-(2-propen-1yl)phenoxy]ethyl ester (CA INDEX NAME)

$$\begin{array}{c} \text{H}_2\text{C} \\ \text{Me} - \text{C} - \text{C} - \text{O} - \text{CH}_2 - \text{CH}_2 - \text{O} \end{array}$$

CC 63-7 (Pharmaceuticals)

IT

Section cross-reference(s): 25, 35 912479-74-4P 912479-75-5P 912479-76-6P

912479-77-79

RL: BSU (Biological study, unclassified); PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(preparation of dental composites and orthopedic bone cements from polymeric derivs. of the natural product eugenol)

375856-97-6P 312479-73-3P

RL: BSU (Biological study, unclassified); RCT (Reactant); SPN

3/13/2008 10/588,210 12 -{ Deleted: 3/6/2008

(Synthetic preparation); BIOL (Biological study); PREP

(Preparation); RACT (Reactant or reagent)

(preparation of dental composites and orthopedic bone cements from

polymeric derivs. of the natural product eugenol)

REFERENCE COUNT:

35 THERE ARE 35 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE

IN THE RE FORMAT

L35 ANSWER 4 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:826263 HCAPLUS Full-text

DOCUMENT NUMBER: 145:438789

TITLE: Part 2: Building Diverse Natural-Product-Like

Architectures from a Tetrahydroaminoquinoline Scaffold. Modular Solution- and Solid-Phase Approaches for Use in High-Throughput Generation

of Chemical Probes

AUTHOR(S): Sharma, Utpal; Srivastava, Stuti; Prakesch,

Michael; Sharma, Maya; Leek, Donald M.; Arya,

Prabhat

CORPORATE SOURCE: Steacie Institute for Molecular Sciences,

National Research Council of Canada, Ottawa, ON,

K1A OR6, Can.

SOURCE: Journal of Combinatorial Chemistry (2006), 8(5),

735-761

CODEN: JCCHFF; ISSN: 1520-4766

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 145:438789

GΙ

RN

AB The solution—and solid—phase synthesis to obtain several natural—product—like, tetrahydroquinoline—based, polycyclic derivs. were developed. In one approach, two derivs. I (R = OAc, R1 = H; R = H, R1 = OAc), having an eight—membered unsatd. lactam, were successfully obtained both in solution and on solid support.

IT 912628-55-8P 312623-59-2P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(solution- and solid-phase synthesis of tetrahydroquinoline-based natural-product-like polycyclic derivs.)

912628-55-8 HCAPLUS

CN Oxazolo[5,4-c]quinoline-1(2H)-carboxylic acid, 4-[(2S)-2-(acetyloxy)-4-pentenyl]-3a,4,5,9b-tetrahydro-8-[(2-methoxyethoxy)methoxy]-2,2-

Deleted: 3/6/2008

dimethyl-5-(1-oxo-2-propenyl)-, phenylmethyl ester, (3as,4s,9bs)-(9CI) (CA INDEX NAME)

Absolute stereochemistry.

RN 912628-59-2 HCAPLUS

CN Oxazolo[5,4-c]quinoline-1(2H)-carboxylic acid, 4-[(2R)-2-(acetyloxy)-4-pentenyl]-3a,4,5,9b-tetrahydro-8-[(2-methoxyethoxy)methoxy]-2,2-dimethyl-5-(1-oxo-2-propenyl)-, phenylmethyl ester, (3aS,4S,9bS)-(9CI) (CA INDEX NAME)

Absolute stereochemistry.

31-5 (Alkaloids) 912628-46-7P 912483-62-6P 912483-77-3P 912628-45-6P 912628-47-8P 912628-50-3P 912628-52-5P 912628-54-7P 912628-55-3P 912628-58-1P 912628-59-2P 912628-61-6DP, resin-bound 912628-61-6P 912628-62-7DP, resin-bound 912628-63-8DP, resin-bound 912628-65-0DP, 912628-65-0P 912628-66-1DP, resin-bound resin-bound 912628-67-2DP, resin-bound 912628-70-7DP, resin-bound 912628-70-7P 912628-72-9P 912628-73-0P 912628-74-1P 912628-75-2P 912628-76-3P 912628-77-4P 912628-78-5P 912628-79-6P 912628-80-9P 912628-81-0P 912628-82-1P 912628-83-2P 912628-84-3P 912628-85-4P 912628-86-5P 912628-87-6P 912628-88-7P 912628-89-8P 912628-90-1P 912628-91-2P 912628-92-3P 912628-93-4P 912628-94-5P 912628-95-6P 912628-96-7P 912628-97-8P 912628-98-9P 912628-99-0P 912629-00-6DP, resin-bound 912629-01-7DP, 912629-02-8DP, resin-bound 912629-03-9DP, resin-bound

3/11/2008 10/588,210 14 Deleted: 3/6/2008

 resin-bound
 912629-04-0DP, resin-bound
 912629-05-1DP,

 resin-bound
 912629-12-0DP, resin-bound
 912629-18-6DP,

 resin-bound
 912629-22-2DP, resin-bound
 912629-24-4DP,

 resin-bound
 912629-25-5DP, resin-bound
 912629-24-4DP,

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(solution- and solid-phase synthesis of tetrahydroquinoline-based

natural-product-like polycyclic derivs.)

REFERENCE COUNT: 25 THERE ARE 25 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L35 ANSWER 5 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:269793 HCAPLUS Full-text

DOCUMENT NUMBER: 144:340792

TITLE: Lithography technique using silicone molds INVENTOR(S): Bahadur, Maneesh; Chen, Wei; Albaugh, John;

Harkness, Brian; Tonge, James PATENT ASSIGNEE(S): Dow Corning Corporation, USA

SOURCE: Dow corning corporation, US

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	TENT I				KIN	D -	DATE			APPL	ICAT	ION :	NO.		D.	ATE
	2006		55		A2		2006	0323		WO 2	005-	US31	150		2	00508
WO	2006	0314	55		А3		2006	1026							J	1
	W: RW:	CH, GB, KP, MW, SC, UG, AT, IE, BF,	CN, GD, KR, MX, SD, US, BE, IS, BJ,	CO, GE, KZ, MZ, SE, UZ, BG, IT, CF,	CR, GH, LC, NA, SG, VC, CH, LT,	CU, GM, LK, NG, SK, VN, CY, LU, CI,	AU, CZ, HR, LR, NI, SL, YU, CZ, LV,	DE, HU, LS, NO, SM, ZA, DE, MC, GA,	DK, ID, LT, NZ, SY, ZM, DK, NL, GN,	DM, IL, LU, OM, TJ, ZW EE, PL,	DZ, IN, LV, PG, TM, ES, PT, GW,	EC, IS, MA, PH, TN, FI, RO, ML,	EE, JP, MD, PL, TR, FR, SE, MR,	EG, KE, MG, PT, TT, GB, SI, NE,	ES, KG, MK, RO, TZ, GR, SK, SN,	FI, KM, MN, RU, UA, HU, TR, TD,
EP	1803	ZW,	,	ΑZ,	BY,	KG,	LS, KZ, 2007	MD,	RU,	ΤJ,	MT	ŕ	·	TZ,	UG,	ZM,
															3	_
	R:						CZ, LU,				,			,		
CN	1010	1907	4		A		2007	0815		CN 2	005-	8003	0625		2	00508 1
US	2007:	2697	47		A1		2007	1122		US 2	007-	6599	89		2	00702
KR	2007	0523	05		A		2007	0521		KR 2	007-	7058	58			00703

Deleted: 3/6/2008

PRIORITY APPLN. INFO.:

US 2004-609425P

200409 13

W

WO 2005-US31150

200508

AB A method includes the steps of: (A) filling a silicone mold having a patterned surface with a curable (meth)acrylate formulation, (B) curing the curable (meth)acrylate formulation to form a patterned feature, (C) separating the silicone mold and the patterned feature, optionally (D) etching the patterned feature, and optionally (E) repeating steps (A) to (D) reusing the silicone mold. The curable (meth)acrylate formulation contains a fluorofunctional (meth)acrylate, a (meth)acrylate, and a photoinitiator.

IT 208995-35-1

RL: NUU (Other use, unclassified); USES (Uses) (lithog. technique using silicone molds)

RN 208995-35-1 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, methyl-2-[methyl-2-(2-propenyloxy)ethoxy]ethyl ester (CA INDEX NAME)

2 (D1_Me)

CC 74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

75-91-2D, 1,1-Dimethylethyl hydroperoxide, reaction product with bis[2,2,6,6-tetramethyl-1-(octyloxy)-4-piperidinyl] ester 78-08-0, Vinyltriethoxysilane 78-10-4, Tetraethoxysilane 79-10-7, Acrylic acid, uses 79-41-4, Methacrylic acid, uses 80-62-6, Methyl methacrylate 86-39-5, 2-Chlorothioxanthone 96-05-9, Allyl methacrylate 97-63-2, Ethyl methacrylate 97-86-9, Isobutyl 97-88-1, n-Butyl methacrylate 97-90-5, Ethylene methacrylate glycol dimethacrylate 97-99-4 101-43-9, Cyclohexyl methacrylate 105-16-8, Diethylaminoethyl methacrylate 106-63-8, Isobutyl acrylate 106-74-1, 2-Ethoxyethyl acrylate 106-91-2, Glycidyl methacrylate 107-98-2, 1-Methoxy-2-propanol 108-46-3, 1,3-Benzenediol, uses 109-16-0, Triethylene glycol dimethacrylate 109-17-1, Tetraethylene glycol dimethacrylate 111-20-6, Decanedioic acid, uses 112-53-8, 1-Dodecanol 119-53-9D, Benzoin, ether 119-61-9, Benzophenone, uses 126-98-7, Methacrylonitrile 142-90-5, Lauryl methacrylate 407-47-6 502-44-3D, Caprolactone, acrylate deriv 585-07-9, tert-Butyl methacrylate 611-73-4 681-84-5, Tetramethoxysilane 688-84-6, 2-Ethylhexyl methacrylate 818-61-1, 2-Hydroxyethyl acrylate 868-77-9, 2-Hydroxyethyl methacrylate 923-26-2, 2-Hydroxypropyl methacrylate 1-Hydroxycyclohexyl phenyl ketone 989-38-8, Rhodamine 6G 999-61-1, 2-Hydroxypropyl acrylate 1070-70-8, 1,4-Butanediol diacrylate 1189-08-8 1330-61-6, Isodecyl acrylate 1663-39-4, tert-Butyl acrylate 2082-81-7, Butanediol dimethacrylate 2156-97-0, Lauryl acrylate 2223-82-7, Neopentyl glycol diacrylate 2358-84-1, Diethylene glycol dimethacrylate 2370-63-0, 2-Ethoxyethyl methacrylate 2399-48-6, Tetrahydrofurfuryl acrylate

Deleted: 3/6/2008

16

2439-35-2, Dimethyl aminoethyl acrylate 2455-24-5, Tetrahydrofurfuryl methacrylate 2461-18-9D, [(Dodecyloxy)methyl]oxirane, reaction product with 4-[4,6-bis(2,4-dimethylphenyl)-1,3,5-triazin-2-yl]4-Hydroxybutyl acrylate 2495-25-2, Tridecyl methacrylate 2495-35-4, Benzyl acrylate 2495-37-6, Benzyl methacrylate 2530-83-8, Glycidoxypropyltrimethoxysilane 2530-85-0 2602-34-8 2768-02-7, Vinyltrimethoxysilane 2867-47-2, Dimethyl aminoethyl methacrylate 3066-71-5, Cyclohexyl acrylate 3076-04-8, Tridecyl acrylate 3121-61-7, 2-Methoxyethyl acrylate 3290-92-4. Trimethylol propane trimethacrylate 3524-68-3, Pentaerythritol triacrylate 4074-88-8, Diethylene glycol diacrylate 4491-03-6, Bisphenol A diacrylate 4813-57-4, Stearyl acrylate 4986-89-4, Pentaerythritol tetraacrylate 5039-78-1 5888-33-5, Isobornyl acrylate 6606-59-3, 1,6-Hexanediol dimethacrylate 6652-28-4, Benzoin isopropyl ether 7328-17-8 7473-98-5, 2-Hydroxy-2-methyl-1-phenylpropan-1-one 7534-94-3, Isobornyl methacrylate 7779-31-9 9016-00-6D, Polydimethylsiloxane, polyether-modified 13048-33-4, 1,6-Hexanediol diacrylate 13402-02-3, Cetyl acrylate 15206-55-0, Methylbenzoyl formate 15625-89-5, Trimethylol propane triacrylate 15895-80-4 17831-71-9, Tetraethylene glycol diacrylate 19485-03-1, 1,3-Butylene glycol diacrylate 21142-29-0, 3-Methacryloxypropyltriethoxysilane 22499-12-3, Benzoin isobutyl 24615-84-7, 2-Carboxyethyl acrylate 24650-42-8, Benzildimethylketal 25154-39-6, Tetrafluoropropyl acrylate 25721-76-0, Polyethylene glycol dimethacrylate 25736-86-1 25736-86-1 25852-49-7, Polypropylene glycol dimethacrylate 26570-48-9, Polyethylene glycol diacrylate 27458-06-6, Benzoylbenzoic acid 27905-45-9 28961-43-5 29570-58-9D, Dipentaerythritol hexaacrylate, caprolactone modified 29590-42-9, Isooctyl acrylate 31621-69-9 31900-57-9, Polydimethylsiloxane 32171-39-4 32360-05-7, Stearyl methacrylate 36811-99-1, 2,2'-(2,5-Thiophendiyl)bis(tert-butylbenzoxazole) 38056-88-1 38785-10-3, Trifluoroethyl methacrylate 39420-45-6, Polypropylene glycol monomethacrylate 39670-09-2 41637-38-1, Bisphenol A ethoxylate dimethacrylate 41680-37-9D, Dipentaerythritol hexamethacrylate, caprolactone modified 42594-17-2, Tricyclodecane dimethanol diacrylate 42978-66-5, Tripropylene glycol diacrylate 48145-04-6, 2-Phenoxyethyl acrylate 51728-26-8 52408-84-1 52408-84-1 53879-54-2 56093-53-9, Pentaerythritol acrylate 57472-68-1, Dipropylene glycol diacrylate 60506-81-2, Dipentaerythritol pentaacrylate 61253-00-7, Octafluoropentyl methacrylate 64111-89-3, Dipropylene glycol dimethacrylate 64401-02-1 67362-76-9, 2-Butoxyethyl-4-dimethylaminobenzoate 73507-02-5, Methyl benzoylbenzoate 75577-70-7, 72829-09-5 Trimethylolpropane ethoxy triacrylate 82799-44-8 83846-85-9, 4-Benzoyl-4'-methyldiphenyl sulfide 84170-74-1 92933-79-4, Octafluoropentyl acrylate 94108-97-1, Ditrimethylolpropane tetraacrylate 119313-12-1 162881-26-7, Phenylbis(2,4,6-trimethylbenzoyl)phosphine oxide 208995-35-1 236422-51-8, Octyl decyl acrylate 880485-38-1 880485-39-2D, reaction product with [(dodecyloxy)methyl]oxirane RL: NUU (Other use, unclassified); USES (Uses)

L35 ANSWER 6 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2005:23389 HCAPLUS Full-text

DOCUMENT NUMBER: 142:115027

TITLE: Production of organic glasses contains

(lithog. technique using silicone molds)

TITLE: Production of organic glasses containing low

3/11/2008

10/588,210

Deleted: 3/6/2008

17

amounts of residual monomers INVENTOR(S):

Arzhakov, M. S.; Arzhakov, S. A.; D'yachkov, A.

I.; D'yachkov, I. A.; Skorobogatova, A. E.;

Stoyachenko, I. L.; Chernavin, V. A.

PATENT ASSIGNEE(S): Russia

SOURCE: Russ., No pp. given

CODEN: RUXXE7

DOCUMENT TYPE: Patent LANGUAGE: Russian

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
RU 2243978	C2	20050110	RU 2001-121094	200107
PRIORITY APPLN. INFO.:			RU 2001-121094	30
				200107 30

A polymer for manufacturing organic glasses is produced by radical polymerization of Me methacrylate or its mixts. with (meth)acrylates or vinyl acetate in the presence of a radical initiator to obtain a polymer-monomer mixture, followed by extrusion with simultaneous depolymn. The radical initiator is a mixture of at least two initiators with different decomposition temps., or an initiator with two different decomposition temps. The method provides polymers containing low amts. of residual monomers (monomer conversions close to 100%). Thus, Me methacrylate was polymerized at 230 $^{\circ}$ in the presence of tert-Bu perbenzoate (0.3) and tert-Bu peroxide (0.1%) to a monomer conversion of 90%, followed by extrusion at 110° with simultaneous depolymn. to a monomer conversion > 99% and a mol. weight of 230,000.

53935-94-7DP, Me methacrylate-based polymers ΙT 58985-96-9DP, Me methacrylate-based polymers RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(production of organic glasses containing low amts. of residual monomers)

RN 58985-94-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-(2-propenyloxy)ethoxy]ethyl ester (9CI) (CA INDEX NAME)

58985-96-9 HCAPLUS RN

2-Propenoic acid, 2-methyl-, 2-[2-[2-(2propenyloxy)ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

$$\begin{array}{c} ^{\rm H2C} \\ ^{\rm Me} - \overset{\circ}{\mathbb{U}} - \overset{\overset{\circ}{\mathbb{U}} - \overset{\circ}{\mathbb{U}} - \overset{\circ}{\mathbb{U}} - \overset{\circ}{\mathbb{U}} - \overset{\circ}{\mathbb{U}} -$$

ICM C08F265-06

ICS C08F002-46; C08J003-28; B29C071-00; B29C055-12

CC 37-3 (Plastics Manufacture and Processing) 80-62-6DP, Methyl methacrylate, polymers 96-05-9DP, Allyl methacrylate, Me methacrylate-based polymers 1025-15-6DP, Triallyl isocyanurate, Me methacrylate-based polymers 2998-04-1DP, Diallyl adipate, Me methacrylate-based polymers 16839-48-8DP, Me methacrylate-based polymers 26330-22-3DP, Me methacrylate-based polymers 26872-73-1DP, Me methacrylate-based polymers 58985-94-709, Me methacrylate-based polymers 58385-96-90P, Me methacrylate-based polymers RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (production of organic glasses containing low amts. of residual monomers) L35 ANSWER 7 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 2004:1041465 HCAPLUS Full-text ACCESSION NUMBER: DOCUMENT NUMBER: 142:177537 Comb-shaped single ion conductors based on polyacrylate ethers and lithium alkyl sulfonate

TITLE:

AUTHOR(S):Sun, Xiao-Guang; Hou, Jun; Kerr, John B. CORPORATE SOURCE: EETD, MS 62-203, Lawrence Berkeley National

Laboratory, Berkeley, CA, 94720, USA

SOURCE: Electrochimica Acta (2005), 50(5), 1139-1147

CODEN: ELCAAV; ISSN: 0013-4686

PUBLISHER: Elsevier B.V. DOCUMENT TYPE: Journal LANGUAGE: English

Comb-shaped single ion conductors have been synthesized by (1) sulfonation of small mol. chloroethyleneglycols, which, after ion exchange to the Li+ salt were then converted to the acrylate by reaction with acryloyl chloride and copolymd. with polyethylene glycol monomethyl ether acrylate (Mn = 454, n = 8) (PAE8-co-E3SO3Li); (2) sulfonation of chloride end groups grafted on to prepolymers of polyacrylate ethers (PAE8-g-EnSO3Li, n = 2, 3). The highest conductivity at 25 $^{\circ}$ C of 2.0 + 10-7 S cm-1 was obtained for the PAE8-co-E3SO3Li with a salt concentration of EO/Li = 40. The conductivity of PAE8-g-E3SO3Li is lower than that of PAE8-co-E3SO3Li at similar salt concns., which is related to the incomplete sulfonation of the grafted polymer that leads to a lower concentration of Li+. The addition of 50 weight% of plasticizer, PC/EMC (1/1, volume/volume), to PAE8-g-E2SO3Li increases the ambient conductivity by three orders of magnitude, which is due to the increased ion mobility in a micro-liquid environment and an increase concentration of free ions as a result of the higher dielec. constant of the solvent. A sym. Li/Li cell with an electrolyte membrane consisting of 75 weight % PC/EMC (1/1, volume/volume) was cycled at a c.d. of 100 μ A cm-2 at 85 °C. The cycling profile showed no concentration polarization after a break-in period during the first few cycles, which was apparently due to reaction of the solvent at the lithium metal surface that reacted with lithium metal to form a stable SEI laver.

\$35628-76-7DP, Diethylene glycol allyl ether acrylate-oxirane graft copolymer, methyl ether, reaction products with $\{3-[2-(2-\text{chloroethoxy})\ \text{ethoxy}]\ \text{propyl}\}-1,1,3,3$ tetramethyldisiloxane and sodium sulfonate, ion-exchanged, lithium RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(comb-shaped single ion conductors based on polyacrylate ethers and lithium alkyl sulfonate)

RN 835628-76-7 HCAPLUS

2-Propenoic acid, 2-[2-(2-propenyloxy)ethoxy]ethyl ester, polymer with oxirane, methyl ether, graft (9CI) (CA INDEX NAME)

H2C = CH - CH2 - O - CH2 - CH2 - O - CH2 - CH2 - O - CH2 - CH2 - CH2

CRN 75-21-8 CMF C2 H4 O

CRN 286834-16-0 CMF C10 H16 O4



CC 37-3 (Plastics Manufacture and Processing) Section cross-reference(s): 52 835628-37-0P 835628-38-1DP, reaction products with allyl ether-containing polyacrylate and sodium sulfonate, ion-exchanged, lithium salts 835628-39-2DP, reaction products with {3-[2-(2-chloroethoxy)ethoxy]propyl}-1,1,3,3-tetramethyldisiloxane and sodium sulfonate, ion-exchanged, lithium salts 835628-74-5P, Ethylene oxide-lithium 2-[2-(2-acryloylethoxy)ethoxy]ethyl sulfonate graft copolymer, methyl ether 835628-76-7DP, Diethylene glycol allyl ether acrylate-oxirane graft copolymer, methyl ether, reaction products with {3-[2-(2-chloroethoxy)ethoxy]propyl}-1,1,3,3tetramethyldisiloxane and sodium sulfonate, ion-exchanged, lithium salts RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (comb-shaped single ion conductors based on polyacrylate ethers and lithium alkyl sulfonate) REFERENCE COUNT: THERE ARE 35 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE

20

Deleted: 3/6/2008

IN THE RE FORMAT

L35 ANSWER 8 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2004:249875 HCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 140:272482

TITLE: Aqueous (meth)acrylate compositions for

water-resistant products with good colorant

dispersion stability

INVENTOR(S): Fukada, Akihiko; Awaji, Toshio; Yoshimune, Soki

PATENT ASSIGNEE(S): Nippon Shokubai Co., Ltd., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 20 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
 JP 2004091757	А	20040325	JP 2002-258847	200209
JP 4052904 PRIORITY APPLN. INFO.:	В2	20080227	JP 2002-258847	04
				200209 04

AB Title compns., useful for jet-printing inks, coatings, adhesives, resists, etc., contain acetal- and/or hemiacetal ester bond-containing (meth)acrylates and optionally photopolymn. initiators. Thus, a reaction product of 2-vinyloxyethoxyethyl methacrylate with tetraethylene glycol was mixed with water, Irgacure 3050 (photopolymn. initiator), and Benzopurpurin 4B (red direct dye) and left for 30 min to show no separation of the dye. The composition was applied on copying paper and UV cured to give a water-resistant coating with no discoloration after 1 min in water.

IT 58985-94-7DF, reaction products with polyols or carboxy-containing compds. 286834-16-0DP, reaction products with polyols or carboxy-containing compds. RL: IMF (Industrial manufacture); RCT (Reactant); TEM (Technical or

RL: IMF (Industrial manufacture); RCT (Reactant); TEM (Technical or engineered material use); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(aqueous compns. containing acetal and/or hemiacetal ester bond-containing (meth)acrylates for water-resistant inks with good colorant dispersion stability)

RN 58985-94-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-(2-propenyloxy)ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 286834-16-0 HCAPLUS

CN 2-Propenoic acid, 2-[2-(2-propenyloxy)ethoxy]ethyl ester (9CI) (CA INDEX NAME)

```
H2C = CH - CH2 - O - CH2 - CH2 - O - CH2 - CH2 - O - CH2 - C
```

IC ICM C08F299-00 ICS C08F020-28

CC 42-12 (Coatings, Inks, and Related Products)

Section cross-reference(s): 37

ΙT 50-70-4DP, D-Sorbitol, reaction products with vinyl ether bond-containing acrylates 112-60-7DP, Tetraethylene glycol, reaction products with vinyl ether bond-containing (meth)acrylates 765-12-8DP, Triethylene glycol divinyl ether, reaction products with triethylene glycol mono(meth)acrylate 818-61-1DP, 2-Hydroxyethyl acrylate, reaction products with triethylene glycol divinyl ether and tetraethylene glycol 9004-75-5DP, Polyethylene glycol monosuccinate, reaction products with vinyl ether bond-containing acrylates 19812-60-3DP, Tetraethylene glycol monoacrylate, reaction products with triethylene glycol unsatd. ethers 21217-75-4DP, Tetraethylene glycol monomethacrylate, reaction products with triethylene glycol unsatd. ethers 25618-55-7DP, Polyglycerine, reaction products with vinyl ether bond-containing acrylates 50586-59-9DP, Polyethylene glycol trimethylolpropane ether, reaction products with vinyl ether bond-containing acrylates 58985-94-7DP, reaction products with polyols or carboxy-containing compds. 76392-22-8DP, reaction products with polyols or carboxy-containing compds. 86273-46-3DP, reaction products with polyols or carboxy-containing compds. 90736-68-8DP, 4,7,10,13-Tetraoxahexadeca-1,15-diene, reaction products with triethylene glycol mono(meth)acrylate 123831-04-9P 286834-16-0DP, reaction products with polyols or carboxy-containing compds. 673477-34-4DP, reaction products with vinyl ether bond-containing acrylates RL: IMF (Industrial manufacture); RCT (Reactant); TEM (Technical or engineered material use); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(aqueous compns. containing acetal and/or hemiacetal ester bond-containing (meth)acrylates for water-resistant inks with good colorant dispersion stability)

L35 ANSWER 9 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2004:158181 HCAPLUS Full-text

DOCUMENT NUMBER: 140:357767

TITLE: Synthesis and characterization of network type

single ion conductors

AUTHOR(S): Sun, Xiao-Guang; Reeder, Craig L.; Kerr, John B. CORPORATE SOURCE: Lawrence Berkeley National Laboratory, Berkeley,

CA, 94720, USA

SOURCE: Macromolecules (2004), 37(6), 2219-2227

CODEN: MAMOBX; ISSN: 0024-9297

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

AB New single ion conductors were synthesized by grafting the allyl group-containing lithium salt, lithium bis(allylmalonato)borate (LiBAMB), onto allyl group-containing comb-branch polyacrylate or polymethacrylate ethers by means of hydrosilylation. The highest ambient temperature conductivity of 3.5 + 10-7 S cm-1 was obtained for a polyacrylate ether-based single ion conductor containing eight EO units in the side chain and five EO units in the crosslinking side chain, to which the anion was fixed with a salt

concentration of EO/Li = 20. For polyacrylate ether-based single ion conductors, an increase of chain length in both side chains and crosslinking anion chains favors an increase of ionic conductivity. The addition of 50 weight % EC/DMC (1/1, weight/weight) increased the ionic conductivity by more than 2 orders of magnitude due to both the increase in ionic mobility from the liquid phase and the increase in the concentration of free ions from the high dielec. constant of the solvent. The preliminary Li/Li cycling profiles of dry polyacrylate—and polymethacrylate ether—based single ion conductors are encouraging as almost no concentration polarization or relaxation was observed. The observed increase in cell potential with cycling is apparently due to an increase in the interfacial impedance associated with the SEI layer, and the cell failure is accompanied by the decomposition of the ester bond of the polyacrylate backbone.

IT 681819-03-4P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(monomer; for preparation of polyacrylate-polymethacrylate-based network-type single ion conductors)

RN 681819-03-4 HCAPLUS

IT 681819-06-7P 681819-10-3P

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(synthesis and characterization of polyacrylate-polymethacrylate-based network-type single ion conductors)

RN 681819-06-7 HCAPLUS

CN 2-Propenoic acid, 2-[2-(2-methoxyethoxy)ethoxy]ethyl ester, polymer with 2-(2-propenyloxy)ethyl 2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 48067-72-7 CMF C10 H18 O5

CM 2

CRN 7784-80-7 CMF C8 H12 O3

RN 681819-10-3 HCAPLUS

CN 2-Propenoic acid, 3,6,9,12,15,18,21,24-octaoxapentacos-1-yl ester, polymer with 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 681819-08-9 CMF C20 H38 O10

PAGE 1-A

MeO-CH2-CH2-O-CH2-CH2-O-

PAGE 1-B

___ CH2__ CH2__ O__ CH2__ CH2__ O__ CH2__ CH2__ O__ С

CM 2

CRN 681819-03-4 CMF C12 H20 O5

H2C = CH - CH2 - O - CH2 - CH2 - CH2 - O - CH2 - C

CC 35-4 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 52

IT 7784-80-7P, 2-Allyloxyethyl acrylate 16839-48-8P, 2-Allyloxyethyl methacrylate 48067-72-7P 51382-35-5P 681819-03-4P 681819-04-5P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(monomer; for preparation of polyacrylate-polymethacrylate-based network-type single ion conductors)

IT 681819-06-72 681819-07-8P 681819-09-0P 681819-10-3P 681819-11-4P 681819-12-5P

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(synthesis and characterization of polyacrylate-polymethacrylate-based network-type single ion conductors)

REFERENCE COUNT:

30 THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L35 ANSWER 10 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2001:842320 HCAPLUS Full-text DOCUMENT NUMBER: 135:376826

10/588,210 Deleted: 3/6/2008

Siloxanyl-containing monomers, their polymers,

and ophthalmic lenses from the polymers

Nakamura, Masataka; Yokota, Mitsuru INVENTOR(S):

PATENT ASSIGNEE(S): Toray Industries, Inc., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
 JP 2001323024	А	20011120	JP 2000-140441	200005
PRIORITY APPLN. INFO.:			JP 2000-140441	200005

OTHER SOURCE(S): MARPAT 135:376826

XO(CH2CHRO)m(CH2)nA (X = group having polymerizable C-C unsatd. bond; R = H, Me; A = siloxanyl; m, n = 2-10), polymers containing the monomers, and ophthalmic lenses made from the polymers are claimed. The lenses such as contact lenses have high O permeability, high water content, and low modulus of elasticity. Contact lenses were manufactured from CH2:CMeCO2(CH2CH2O)2(CH2)3Si(OSiMe3)3 (preparation given), N,Ndimethylacrylamide, and triethylene glycol dimethacrylate.

58985-94-7P TT

> RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation of siloxanyl-containing monomers, their polymers, and contact lenses therefrom)

58985-94-7 HCAPLUS RN

2-Propenoic acid, 2-methyl-, 2-[2-(2-propenyloxy)ethoxy]ethyl ester CN (9CI) (CA INDEX NAME)

TC TCM C08F030-08

ICS A61L027-00; C07F007-08; C08F299-08; G02B001-04; G02C007-04

63-7 (Pharmaceuticals) CC

Section cross-reference(s): 38

15075-50-0P, Diethylene glycol monoallyl ether 58985-94-7P

374534-72-2P 374534-73-3P 374534-74-4P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);

RACT (Reactant or reagent)

(preparation of siloxanyl-containing monomers, their polymers, and contact lenses therefrom)

L35 ANSWER 11 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2001:780791 HCAPLUS Full-text

DOCUMENT NUMBER: 135:319624

TITLE: Compositions for three-dimensional printing of

solid objects

3/11/2008 10/588,210 25 \[\sqrt{Deleted: 3/6/2008}

INVENTOR(S):

Bredt, James F.; Clark, Sarah L.; Uy, Evert F.;

Dicologero, Matthew J.; Anderson, Timothy;

Tarkenian, Michael Z Corporation, USA

PATENT ASSIGNEE(S): SOURCE:

PCT Int. Appl., 34 pp.

CODEN: PIXXD2
DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

	TENT NO.		KIND	DATE	APPLICATION NO.	DATE
	 2001078969)	A2	20011025	WO 2001-US12220	200104
WO	2001078969 W: CA, 3		A3	20020530		13
		BE, CH, PT, SE,		E, DK, ES,	FI, FR, GB, GR, IE, IT,	LU, MC,
CA	2405539			20011025	CA 2001-2405539	200104 13
US	2001050031	L	A1	20011213	US 2001-835292	
						200104 13
EP	1272334		A2	20030108	EP 2001-927008	200104 13
					GB, GR, IT, LI, LU, NL, MK, CY, AL, TR	SE, MC,
JP					JP 2001-576254	200104 13
US	2005197431	L	A1	20050908	US 2005-68487	200502 28
PRIORIT	Y APPLN. IN	NFO.:			US 2000-197118P F	200004
					US 2000-197526P F	200004 14
					US 2001-835292 H	200104 13
					WO 2001-US12220 V	v 200104 13

AB The composition for selectively adhering particulate material to form a solid object in a three-dimensional printer comprises a nonaq. organic compound

IT 208995-35-1, Di(propylene glycol) allyl ether methacrylate RL: TEM (Technical or engineered material use); USES (Uses) (compns. for three-dimensional printing of solid objects)

RN 208995-35-1 HCAPLUS

8/11/2008 10/588,210

Deleted: 3/6/2008

26

CN 2-Propenoic acid, 2-methyl-, methyl-2-[methyl-2-(2propenyloxy)ethoxy]ethyl ester (CA INDEX NAME)

2 (D1_Me)

ICM B29C067-00 ICS B29C041-00; B41J002-01; C09D011-10 CC42-11 (Coatings, Inks, and Related Products) 64-17-5, Ethanol, uses 67-63-0, Isopropanol, uses 75-65-0. tert-Butanol, uses 97-88-1, Butyl methacrylate 106-65-0, Dimethyl succinate 111-55-7, Ethylene glycol diacetate 123-25-1, Diethyl succinate 141-78-6, Ethyl acetate, uses 471-34-1, Calcium carbonate, uses 544-17-2, Calcium formate 627-93-0, Dimethyl adipate 1189-08-8, 1,3-Butylene glycol dimethacrylate 1305-62-0, Calcium hydroxide, uses 1314-13-2, Zinc oxide, uses 1317-61-9, Iron oxide (Fe304), uses 1327-44-2, Potassium aluminum silicate 1330-43-4, Sodium tetraborate 1335-30-4, Aluminum silicate 1344-09-8, Sodium silicate 1344-28-1, Aluminum oxide, uses 1344-95-2, Calcium silicate 1985-51-9, Neopentyl glycol dimethacrylate 6484-52-2, Ammonium nitrate, uses 6606-59-3, 1,6-Hexanediol dimethacrylate 7558-79-4 7631-86-9, Silica, uses 7647-14-5, Sodium chloride, uses 7778-80-5, Potassium sulfate, uses 7783-28-0, Ammonium hydrogen phosphate 9002-89-5, Polyvinyl alcohol 9002-98-6 9003-20-7, Polyvinyl acetate 9003-39-8, Polyvinyl pyrrolidone 9003-53-6, Polystyrene 9003-56-9, ABS 9011-14-7, Polymethyl methacrylate 9080-79-9, Sodium polystyrene sulfonate 10042-91-8 11104-48-6, Calcium aluminate 12125-02-9, Ammonium chloride, uses 13048-33-4, 1,6-Hexanediol diacrylate 13463-67-7, Titania, uses 18023-33-1, Vinyltriisopropoxysilane 25086-89-9, Vinyl acetate-vinyl pyrrolidone copolymer 25087-26-7, Polymethacrylic acid 26062-79-3, Polydiallyldimethylammonium chloride 26124-23-2, Vinylpyrrolidone-acrylamide copolymer 31113-94-7, Vinyl methyl ether-vinyl pyrrolidone copolymer 42978-66-5, Tri(propylene glycol) diacrylate 48145-04-6, Ethylene glycol phenyl ether acrylate 54193-36-1, Polymethacrylic acid sodium salt 208995-35-1, Di(propylene glycol) allyl ether methacrylate 367277-91-6, Vinylpyrrolidone-2-ethyl-2-oxazoline

RL: TEM (Technical or engineered material use); USES (Uses) (compns. for three-dimensional printing of solid objects)

L35 ANSWER 12 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1998:600088 HCAPLUS Full-text

DOCUMENT NUMBER: 129:291215

TITLE: Curable acrylic polysiloxane compositions for

flexible weather-resistant coatings

INVENTOR(S): Iga, Nobuo; Oosugi, Koji

PATENT ASSIGNEE(S): Nippon Paint Co., Ltd., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

copolymer

Deleted: 3/6/2008

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
 JP 10245488	A	19980914	JP 1997-65402	199703
PRIORITY APPLN. INFO.:			JP 1997-65402	03
				199703 03

The compns. comprise (a) polysiloxanes bearing ≥2 SiH groups/mol., (b) alkenyl group-containing acrylic polymers (iodine value 40-100, Mn 1000-100,000), and (c) hydrosilylation catalysts, where the mol. ratio of SiH in (a) to alkenyl in (b) is 0.5-4. Thus, a varnish of 2-allyloxyethyl methacrylate-styrene-methacrylic acid-cyclohexyl methacrylate copolymer (Mn 4100, iodine value 60) was mixed with di-Ph Me H polysiloxane, 2% EtOH solution of chloroplatinic acid, and 2-methyl-3-butyn-2-ol, applied on an Fe sheet, and baked to give clear coatings showing excellent acid resistance, pencil hardness H, and good resistance to xylene rubbing test.

IT 214133-80-90F, reaction products with hydrogen polysiloxanes 214133-81-00F, reaction products with hydrogen polysiloxanes 214133-83-20F, reaction products with hydrogen polysiloxanes 214133-84-30F, reaction products with hydrogen polysiloxanes RL: IMF (Industrial manufacture); PRF (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (curable acrylic polysiloxane coatings with excellent weather resistance and flexibility)

RN 214133-80-9 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, cyclohexyl ester, polymer with ethenylbenzene, methyl 2-methyl-2-propenoate and 2-[2-(2-propenyloxy)ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CF INDEX NAME)

CM 1

CRN 58985-94-7 CMF C11 H18 O4

CM 2

CRN 101-43-9 CMF C10 H16 O2

$$\text{Out} \subseteq \text{CH}_2$$

CM 3

CRN 100-42-5 CMF C8 H8

H 2 C - CH - Ph

CM 4

CRN 80-62-6 CMF C5 H8 O2



RN 214133-81-0 HCAPLUS

2-Propenoic acid, 2-methyl-, cyclohexyl ester, polymer with ethenylbenzene, ethyl 2-propenoate, methyl 2-methyl-2-propenoate and 2-[2-(2-propenyloxy)ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-94-7 CMF C11 H18 O4

$$\begin{array}{c} ^{\text{H2C}} \overset{\circ}{\text{Me}} \overset{\circ}{\text{U}} \overset{\circ}{\text{U}} \overset{\circ}{\text{U}} \overset{\circ}{\text{O}} \overset{\circ}{\text{CH}_2} \overset{\circ}{\text{CH}_2$$

CM 2

CRN 140-88-5 CMF C5 H8 O2

CM 3

CRN 101-43-9 CMF C10 H16 O2

CM 4

CRN 100-42-5 CMF C8 H8

CM 5

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{c} {}^{\text{H}2\text{C}} \\ {}^{\text{Me}} = {}^{\text{C}} = {}^{\text{C}} = {}^{\text{OMe}} \end{array}$$

RN 214133-83-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, cyclohexyl ester, polymer with ethenylbenzene, N-(1-methylethyl)-2-propenamide, methyl 2-methyl-2-propenoate and 2-[2-(2-propenyloxy)ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

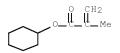
CRN 58985-94-7 CMF C11 H18 O4

CM 2

CRN 2210-25-5 CMF C6 H11 N O

CM 3

CRN 101-43-9 CMF C10 H16 O2



CM 4

CRN 100-42-5 CMF C8 H8

CM 5

CRN 80-62-6 CMF C5 H8 O2

RN 214133-84-3 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, cyclohexyl ester, polymer with ethenylbenzene, ethyl 2-propenoate, N-(1-methylethyl)-2-propenamide, methyl 2-methyl-2-propenoate and 2-[2-(2-propenyloxy)ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-94-7 CMF C11 H18 O4

$$\begin{array}{c} \text{H2C} & \circ \\ \parallel & \parallel \\ \text{Me-C-C-O-CH}_2 - \text{CH}_2 - \text{CH}_2$$

CM 2

CRN 2210-25-5

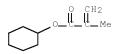
CMF C6 H11 N O

CM 3

CRN 140-88-5 CMF C5 H8 O2

CM 4

CRN 101-43-9 CMF C10 H16 O2



CM 5

CRN 100-42-5 CMF C8 H8

CM 6

CRN 80-62-6 CMF C5 H8 O2

ICM C08L083-05 ICS C08G081-02; C08L029-10; C08L033-06; C08L033-14; C09D129-10; C09D133-06; C09D133-14; C09D183-05

42-7 (Coatings, Inks, and Related Products) ТТ 155904-19-1DP, Diphenylsilanediol-methylsilanediol copolymer,

trimethylsilyl-terminated, reaction products with alkenyl-containing acrylic polymers 214133-64-9DP, 2-Allyloxyethyl

methacrylate-cyclohexyl methacrylate-methyl methacrylate-styrene copolymer, reaction products with hydrogen polysiloxanes

214133-65-0DP, reaction products with hydrogen polysiloxanes

214133-66-1DP, reaction products with hydrogen polysiloxanes

214133-67-2DP, reaction products with hydrogen polysiloxanes 214133-69-4DP, 2-Allyloxyethyl acrylate-cyclohexyl

methacrylate-methyl methacrylate-styrene copolymer, reaction products with hydrogen polysiloxanes 214133-70-7DP, reaction 214133-71-8DP, reaction products with hydrogen polysiloxanes

products with hydrogen polysiloxanes 214133-72-9DP, reaction 214133-74-1DP, Cyclohexyl products with di-Ph Me H polysiloxane methacrylate-9-decenyl methacrylate-methyl methacrylate-styrene

copolymer, reaction products with hydrogen polysiloxanes

214133-75-2DP, reaction products with hydrogen polysiloxanes

214133-76-3DP, reaction products with hydrogen polysiloxanes

214133-79-6DP, reaction products with hydrogen polysiloxanes

214133-80-950, reaction products with hydrogen polysiloxanes 214133-81-050, reaction products with hydrogen polysiloxanes

214133-83-2DP, reaction products with hydrogen polysiloxanes

214133-84-3DP, reaction products with hydrogen polysiloxanes RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical

or engineered material use); PREP (Preparation); USES (Uses) (curable acrylic polysiloxane coatings with excellent weather

resistance and flexibility)

L35 ANSWER 13 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 1998:397820 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 129:68138

Preparation of alkoxylated glycidyl TITLE:

(meth)acrylates

INVENTOR(S): Fan, Mingxin; Ceska, Gary W.; Horgan, James;

Hazell, Thomas W. Sartomer Co., USA

PATENT ASSIGNEE(S): SOURCE: U.S., 4 pp.

CODEN: USXXAM DOCUMENT TYPE: Patent

English LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

Ε	PA:	ENT :	NO.			KIN	D DA	Ë	AI	PPLIC.	ATION	NO.		D.	ATE
-															
Į	JS	5763	- 629			Α	199	80609	US	3 199	6-772	979			
														_	99612
V	οĪο	9828	287			A 1	190	80702	W	199	7-EP7:	283		2	3
٠	•	3020	20,			111	10.	,00,102	***	, 1,,	, п.	505		1	99712
														2	2
		W:	CA,	CN,	JP,	KR,	MX, PI	J, US							
		RW:	,	BE, SE	CH,	DE,	DK, ES	5, FI,	FR, (GB, G	R, IE,	IT,	LU,	MC,	NL,
Ε	ΞP	9465	,			A1	199	91006	E	199	7-954	168			
														1 2	99712

R: BE, DE, ES, FR, GB, IT, NL

PRIORITY APPLN. INFO.:

US 1996-772979

199612

23

WO 1997-EP7283

199712 2.2

OTHER SOURCE(S): MARPAT 129:68138

Alkoxylated glycidyl (meth)acrylates are prepared by epoxidizing alkoxylated allyl (meth)acrylates with H2O2 in the presence of (a) tungstic acid or its metal salts, (b) phosphoric acid or its metal salts, and (c) ≥1 phase transfer catalyst. Thus, 100.0 g propoxylated allyl methacrylate prepared from propoxylated allyl alc. and methacrylic acid, was mixed with 2.7 g MeN[(CH2)7CH3]3 PO4[WO(O2)2]4 in 100 g toluene, and 50 mL H2O2 (30%) was added into the mixture in 30 min at 60° and reacted for 22.0 h to yield a epoxide at conversion 85%.

208995-35-1P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(preparation of alkoxylated glycidyl (meth)acrylates)

208995-35-1 HCAPLUS RN

2-Propenoic acid, 2-methyl-, methyl-2-[methyl-2-(2-CN propenyloxy)ethoxy]ethyl ester (CA INDEX NAME)

2 (D1_Me)

ICM C07D301-12

INCL 549531000

35-2 (Chemistry of Synthetic High Polymers)

208995-34-0P 208995-35-1P TT

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent)

(preparation of alkoxylated glycidyl (meth)acrylates)

THERE ARE 2 CITED REFERENCES AVAILABLE FOR REFERENCE COUNT: 2

THIS RECORD. ALL CITATIONS AVAILABLE IN

THE RE FORMAT

L35 ANSWER 14 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1995:737303 HCAPLUS Full-text

DOCUMENT NUMBER: 123:113219

TITLE: Silicon-modified acrylic copolymer and adhesive

INVENTOR(S): Hosoi, Yasuhiro; Iwamoto, Osamu; Himeno,

Masataka

PATENT ASSIGNEE(S): Tokuyama Corp., Japan SOURCE: Eur. Pat. Appl., 64 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PA:	TENT NO.	KIND	DATE	APPLICATION NO.	DATE
 EP	632060	A2	19950104	EP 1994-304825	199407 01
EP	632060 632060 R: DE, FR, GB	A3 B1	19950125 19981014		01
	07070246	А	19950314	JP 1994-67362	199404 05
	3105733 07076611		20001106 19950320	JP 1994-91902	199404
	3107702 5476912	В2 А	20001113 19951219	US 1994-268321	28
PRIORITY	Y APPLN. INFO.:			JP 1993-164640	199406 30 A 199307
					02
				JP 1993-175176 i	199307 15
				JP 1994-67362	A 199404 05
				JP 1994-91902 i	199404 28

AB A silicone-modified acrylic copolymer having a weight-average mol. weight (Mw) 5000-1,000,000, comprising (a) acrylate structural unit having an C<13-alkyl group or an aryl group having 6-14 C atoms; (b) a siloxane modified acrylate unit where siloxane is covalently bound through A which is a divalent organic group having 2-20 \mbox{C} atoms which may optionally comprise an ether bond or an ester bond in the main chain; optionally (c) acrylate structural unit having an ethylenically unsatd. hydrocarbon group of 2-20 C atoms which may optionally comprise an ether bond or an ester in the main chain with a:b:c (10-99.9):(90-0.1):(0-89.9). An adhesive was prepared from the graft polymer (Mw 160,000) of Me3Sio(Me2Sio)10(MeHSio)10(MePhSio)10S iMe3 27.9, catalyst 0.33, and ally methacrylate-Me methacrylate copolymer 5 g, dissolved in CH2Cl2, applied on acrylic plate, and bonded with a silicone paste, showing adhesion (to acrylic plate even after 3 min warm water soaking) >20 Kg/cm2. 53935-96-9DP, graft polymer with SiH group-containing siloxane RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (preparation and adhesion properties of)

RN 58985-96-9 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-CH2-CH2-CH2-CH2

ICM C08F008-42 IC ICS C08G081-02

35-8 (Chemistry of Synthetic High Polymers) Section cross-reference(s): 38, 63

80-62-6DP, Methyl methacrylate, graft polymer with SiH group-containing siloxane 96-05-9DP, Allyl methacrylate, graft polymer with SiH group-containing siloxane 96-33-3DP, Methyl acrylate, graft polymer with SiH group-containing siloxane 999-55-3DP, Allyl acrylate, graft polymer with SiH group-containing siloxane 2210-28-8DP, Propyl methacrylate, graft polymer with SiH group-containing siloxane 2495-25-2DP, Tridecyl methacrylate, graft polymer with SiH group-containing siloxane 2495-37-6DP, Benzyl methacrylate, graft polymer with SiH group-containing siloxane 4245-37-8DP, Vinyl methacrylate, graft polymer with SiH group-containing siloxane 13533-08-9DP, graft polymer with SiH group-containing siloxane 58935-96-9DP, graft polymer with SiH group-containing siloxane 110083-27-7DP, graft polymer with SiH group-containing siloxane RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (preparation and adhesion properties of)

L35 ANSWER 15 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1991:196399 HCAPLUS Full-text

DOCUMENT NUMBER: 114:196399

TITLE: Waterless presensitized lithographic plate with photosensitive layer containing allyloxyethyl

acrylate copolymer

INVENTOR(S): Azuma, Tatsuji; Kawamura, Koichi; Kita, Nobuyuki

PATENT ASSIGNEE(S): Fuji Photo Film Co., Ltd., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 11 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
 JP 02226248	А	19900907	JP 1989-46955	198902
PRIORITY APPLN. INFO.:			JP 1989-46955	28
				198902 28

The title plate comprises a substrate coated with a photosensitive layer containing (1) a copolymer having ≥30 mol% structural unit CH2CRCO2(CH2CHR10)nCH2CH:CH2 (I; R, R1 = H, Me; n = 1-15); (2) a monomer or oligomer having ≥1 photopolymerizable ethylenic double bond; and (3) a photopolymn. initiator, and a silicone rubber layer. The plate provides high-quality images and shows good printing durability. Thus, an Al plate with a primer layer was coated with a composition containing 2-methacryloxyethylhydrogen succinate-I (R = Me, R1 = H, n = 1) copolymer, [CH:CMeCO2CH2CH(OH)CH2OCH2]2CHOH, and II, and with a silicone rubber layer to give a presensitized lithog. plate.

133411-78-6, 2-Allyloxyethyl methacrylate-methacrylic acid-triethyleneglycol monomethacrylate copolymer sodium salt RL: USES (Uses)

(lithog. plate photosensitive layer using)

RN 133411-78-6 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with 2-[2-(2-hydroxyethoxy)ethoxy]ethyl 2-methyl-2-propenoate and 2-(2-propenyloxy)ethyl 2-methyl-2-propenoate, sodium salt (9CI) (CA INDEX NAME)

CM 1

CRN 133411-77-5 CMF (C10 H18 O5 . C9 H14 O3 . C4 H6 O2)x CCI PMS CM 2 CRN 16839-48-8

C9 H14 O3

CM 3

CMF

CRN 2351-42-0 CMF C10 H18 O5

4 CM

CRN 79-41-4 CMF C4 H6 O2

ICM G03F007-00

ICS G03F007-038

CC 74-6 (Radiation Chemistry, Photochemistry, and Photographic and

Other Reprographic Processes)

26570-48-9 86829-42-7 133394-54-4, 2-Allyloxyethyl

methacrylate-2-methacryloxyethylhydrogen succinate copolymer

133394-55-5, 2-Allyloxyethyl methacrylate-methacrylic acid copolymer

 $133394-56-6,\ 2-\texttt{Acryloxyethylhydrogen}\ succinate-2-\texttt{allyloxyethyl}$

 ${\tt methacrylate-2-hydroxyethyl\ methacrylate\ copolymer}$ 133411-78-6, 2-Allyloxyethyl methacrylate-methacrylic

acid-triethyleneglycol monomethacrylate copolymer sodium salt

133536-76-2

RL: USES (Uses)

(lithog. plate photosensitive layer using)

L35 ANSWER 16 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 1988:152251 HCAPLUS Full-text

ACCESSION NUMBER:

DOCUMENT NUMBER: 108:152251

TITLE: Urethane (meth)acrylates for coating materials Fukuchi, Shuzo; Yamaguchi, Shigeru INVENTOR(S):

PATENT ASSIGNEE (S): Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan

Jpn. Kokai Tokkyo Koho, 15 pp. SOURCE:

CODEN: JKXXAF DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
 JP 62230762	A	19871009	JP 1986-70883	198603
PRIORITY APPLN. INFO.:			JP 1986- 7 0883	31 198603
				31

Urethane (meth) acrylates are prepared by reacting hydroxyalkyl (meth) acrylates with (un)substituted alkenyl glycidyl ethers and optionally other cyclic compds. and treating with other hydroxy compds. and polyisocyanates. Thus, 2hydroxyethyl acrylate was treated with allyl glycidyl ether at ratio 1:3 to qive CH2:CHC02CH2CH2O [CH2CH(CH2OCH2CH:CH2)0]3H, treated with 2,4-toluene diisocyanate to give a urethane acrylate, treated with 2-hydroxyethyl acrylate, mixed with pentaerythritol tetra(3-mercaptopropionate) and Irgacure 651, coated on steel, and irradiated with a high-pressure Hg lamp to form a coating having pencil hardness 5H.

112861-63-9P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent) (manufacture and reaction of, with polyisocyanates)

RN 112861-63-9 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-hydroxymethylethoxy)methylethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

2 (D1-Me)

IT 112861-61-7P 112861-62-3P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (manufacture and reaction of, with toluene diisocyanate)

RN 112861-61-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-[2-[2-hydroxy](2propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 112861-62-8 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-hydroxy[(2propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

IC ICM C07C125-06

ICS C08G018-67

ICA C08F299-06

CC 42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 55

IT 112861-63-9P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with polyisocyanates)

IT 112861-61-7P 112861-62-8P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with toluene diisocyanate)

L35 ANSWER 17 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1988:96319 HCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 108:96319

TITLE: Light-curable polyene-polythiol coating

materials

INVENTOR(S): Fukuchi, Shuzo; Yamaguchi, Shigeru

PATENT ASSIGNEE(S): Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PRI

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
 JP 62241925	А	19871022	JP 1986-84328	
				198604 14
IORITY APPLN. INFO.:			JP 1986-84328	
				198604
				14

AB Coating materials contain compds. having ≥2 SH groups/mol. and alkenyl group-containing urethane (meth)acrylates. The reaction of 2-hydroxyethyl methacrylate with allyl glycidyl ether gave CH2:CMeCO2CH2CH2O[CH2CH2CH2CH2CH2CH2O]3H which was treated with 2,4-TDI to give a urethane methacrylate, mixed (60 parts) with 40 parts pentaerythritol tetra(3-mercaptopropenoate), coated on steel, and irradiated with high-pressure Hg lamp to form a coating having pencil hardness 5 H.

IT 112961-62-3P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with toluene diisocyanate)

RN 112861-62-8 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-hydroxy](2propenyloxy)methyllethoxyl[(2-propenyloxy)methyl]

propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

IT 112861-61-7P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent)
(manufacture and reaction of, with toluene diisocyanate and trimethylhexamethylene diisocyanate)

RN 112861-61-7 HCAPLUS

2-Propenoic acid, 2-methyl-, 2-[2-[2-[2-hydroxy](2-propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

IT 1.12861-63-99

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (manufacture and reaction of, with trimethylhexamethylene diisocyanate)

RN 112861-63-9 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-hydroxymethylethoxy)methylethoxy][(2-propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

$$H_2C \longrightarrow CH - CH_2 - O - CH_2 - D1$$

2 (D1-Me)

IC ICM C08G075-04

CC 42-10 (Coatings, Inks, and Related Products)
 Section cross-reference(s): 55

IT 112361-62-3P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with toluene diisocyanate)

IT 112861-61-7P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with toluene diisocyanate and trimethylhexamethylene diisocyanate)

IT 112861-63-9F

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with trimethylhexamethylene ${\tt diisocyanate}$)

L35 ANSWER 18 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1988:96318 HCAPLUS Full-text

10/588,210

41

Deleted: 3/6/2008

DOCUMENT NUMBER:

108:96318

Light curable polyene-polythiol coating

materials

INVENTOR(S):

TITLE:

Fukuchi, Shuzo; Yamaguchi, Shigeru

PATENT ASSIGNEE(S): Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent Japanese LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 62241924	A	19871022	JP 1986-84327	
				198604
				14

PRIORITY APPLN. INFO.: JP 1986-84327

198604

14

GT



Coating materials contain compds. having ≥2 SH groups/mol. in alkenyl group-AB containing (meth)acrylates CH2:CRCO2ZOAlBmH (R = H or Me, Z = C2-8 divalent organic groups, A = ring-opened groups of alkenyl glycidyl ethers I, B = ringopened groups of cyclic compds. except I, l = 1-20, m = 0 or 1-20, the arrangement of A and B being arbitrary), and R1 = H or <C20 (halo substituted) hydrocarbyl groups and R2 <C20 alkenyl groups in I. The reaction of 2hydroxyethyl acrylate with allyl glycidyl ether with 98.1% CH2:CHCO2CH2CH2O[CH2CH(CH2OCH2CH:CH2)O]3H which (56 parts) was mixed with 44 parts pentaerythritol tetra(3-mercaptopropenoate), coated on steel, and irradiated with high-pressure Hg damp to form a coating having pencil hardness 3 н.

112861-61-7P 112861-62-8P 112861-63-9P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent)

(manufacture and photochem. polymerization of, with polydiols, for coatings)

112861-61-7 HCAPLUS

2-Propenoic acid, 2-methyl-, 2-[2-[2-[2-hydroxy](2propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 112861-62-8 HCAPLUS

J 2-Propenoic acid, 2-[2-[2-[2-hydroxy[(2propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 112861-63-9 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-hydroxymethylethoxy)methylethoxy][(2-propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

$$H_2C$$
 \longrightarrow CH \longrightarrow CH_2 \longrightarrow O \longrightarrow CH_2 \longrightarrow $D1$

IC ICM C08G075-04

CC $\,$ 42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 55

IT 112861-61-7*P* 112861-62-8P 112861-63-9P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent)

(manufacture and photochem. polymerization of, with polydiols, for coatings)

L35 ANSWER 19 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1986:412173 HCAPLUS Full-text

DOCUMENT NUMBER: 105:12173

ORIGINAL REFERENCE NO.: 105:2033a,2036a

TITLE: Curable resin compositions for dental use

INVENTOR(S): Sakashita, Takeshi; Nakano, Takayuki

PATENT ASSIGNEE(S): Mitsui Petrochemical Industries, Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent

10/588,210

43

Deleted: 3/6/2008

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.		KIND	DATE	APPLICATION NO.	DATE
 JP 61036204	_	Α	19860220	JP 1984-155621	198407
JP 03078841 PRIORITY APPLN. INF	0.:	В	19911217	JP 1984-155621	27
					198407 27

GΙ

AB Curable resin compns. for dental use comprise (1) poly(meth)acryloyloxyalkyl(iso)cyanurate I or II (R1-R3 = H, alkyl, acryloyl, methacryloyl; X1-X3 = oxyalkylene, polyoxyalkylene), (2) vinyl compds., and (3) a polymerization initiator. Thus, bis(methacryloyloxyethyl)(hydroxyethyl)isocyanurate (1 g), a vinyl compound (1 g), camphorquinone (6 mg), Me N,N-dimethylanthranilate (6 mg), and powdered silica (2 g) were mixed to form a paste, which was irradiated with visible light (350-700 nm) for 30 s. The Brinell hardness was 70 and the bending strength was 1150 kg/cm2.

102770-35-4

RL: BIOL (Biological study)

(dental composites or cements containing)

102770-35-4 HCAPLUS RN

2-Propenoic acid, 2-methyl-, 2-[2-[(3-methyl-3butenyl)oxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

ICM A61K006-08 IC

ICA C08F220-40; C08F226-06

CC 63-7 (Pharmaceuticals)

35838-12-1 109-16-0 1565-94-2 3077-12-1 56745-15-4

82508-13-2 102770-35-4

RL: BIOL (Biological study)

(dental composites or cements containing)

L35 ANSWER 20 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 1985:542411 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 103:142411

ORIGINAL REFERENCE NO.: 103:22829a,22832a

TITLE: Synthesis of polymers containing crown lactone

units via cyclopolymerization in the presence of

alkylaluminum chlorides

AUTHOR(S): Yokota, Kazuaki; Kakuchi, Toyoji; Taniguchi,

Yasuyuki; Takada, Yoshiyuki

CORPORATE SOURCE: Fac. Eng., Hokkaido Univ., Sapporo, 060, Japan SOURCE: Makromolekulare Chemie, Rapid Communications

(1985), 6(3), 155-61

CODEN: MCRCD4; ISSN: 0173-2803

DOCUMENT TYPE: Journal LANGUAGE: English

AB Polymerization of the acrylates CH2:CHCH2(OCH2CH2)yO-o-C6H4O(CH2CH2O)xCOCH:CH2 (x=1-3, y=1-2) or o-CH2:CHCH2C6H4O(CH2CH2O)xCOCH:CH2 (x=2-5) by Al chloroalkyls gave

cyclopolymers containing crown ether lactone units. Et3Al2Cl3 and EtAlCl2 were more effective than EtAlCl, but often gave insol. polymers. The polymns. were faster than radical cyclopolymn. In the extraction of alkali metal picrates the cation binding ability of the crown ether lactone derivative polymers decreased in the order: benzo-21-crown > benzo-24-crown-7 benzo-23-crown-6 > benzo-18-crown-5 > benzo-20-crown-5 > benzo-15-crown-4, benzo-14-crown-3, benzo-17-crown-4. For other crown ether polymers, stability was also greatest for rings containing 6 O atoms.

IT 77504-04-2P 77504-06-4P 77504-08-6P 96387-33-6P 96387-35-8P 96387-37-0P

96387-39-2P

RL: SPN (Synthetic preparation); PREP (Preparation)

(crown ether lactone-containing, preparation and metal binding capacity of)

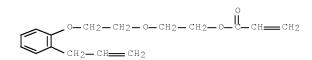
RN 77504-04-2 HCAPLUS

2-Propenoic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CN

CRN 77504-03-1 CMF C16 H20 O4



RN 77504-06-4 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 77504-05-3 CMF C18 H24 O5

RN 77504-08-6 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 77504-07-5 CMF C20 H28 O6

PAGE 1-B

—CH2

RN 96387-33-6 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-propenyloxy)ethoxy]phenoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 96387-32-5 CMF C16 H20 O5

96387-35-8 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-(2-propenyloxy)ethoxy]phenoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM I

RN

CRN 96387-34-7

CMF C18 H24 O6

RN 96387-37-0 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]phenoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 96387-36-9 CMF C20 H28 O7

RN 96387-39-2 HCAPLUS

N 2-Propenoic acid, 2-[2-[2-[2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]phen oxy]ethoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 96387-38-1 CMF C22 H32 O8

IT 96387-32-5P 96387-34-7P 96387-36-9P

96387-38-1P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN 96387-32-5 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-propenyloxy)ethoxy]phenoxy]ethyl ester (9CI) (CA INDEX NAME)

96387-34-7 HCAPLUS

 $\hbox{2-Propenoic acid, 2-[2-[2-(2-propenyloxy)ethoxy]phenoxy]ethoxy]et}\\$ CN hyl ester (9CI) (CA INDEX NAME)

RN96387-36-9 HCAPLUS

2-Propenoic acid, 2-[2-[2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]phenoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

96387-38-1 HCAPLUS RN

2-Propenoic acid, 2-[2-[2-[2-[2-(2-propenyloxy)]]] ethoxy]phen oxy]ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

35-3 (Chemistry of Synthetic High Polymers) 77504-04-2P 77504-06-4P 77504-08-6P CC

ΙT

96387-33-6P 96387-35-3P 96387-37-0P

95387-39-2P 96387-41-6P

RL: SPN (Synthetic preparation); PREP (Preparation)

(crown ether lactone-containing, preparation and metal binding capacity of)

96387-32-5P 96387-34-7P 96387-36-9P

96337-38-1P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

L35 ANSWER 21 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1984:438883 HCAPLUS Full-text

DOCUMENT NUMBER: 101:38883 ORIGINAL REFERENCE NO.: 101:6095a,6098a

Studies on cyclopolymerization in the presence TITLE:

of alkylaluminum chlorides. VII. Cyclopolymerizations of acrylates and

methacrylates containing oligooxyethylene units

in the 11-20-membered-ring region

AUTHOR(S):Yokota, Kazuaki; Kakuchi, Toyoji; Iiyama,

Takashi; Takada, Yoshiyuki

CORPORATE SOURCE: Fac. Eng., Hokkaido Univ., Sapporo, 060, Japan SOURCE: Polymer Journal (Tokyo, Japan) (1984), 16(2),

145-50

CODEN: POLJB8; ISSN: 0032-3896

DOCUMENT TYPE: Journal LANGUAGE: English

An investigation of cyclopolymn. in the presence of alkylaluminum chlorides was extended to 2-[2-(o-allylphenoxy)ethoxy]ethyl acrylate [77504-03-1] and higher homologs containing oligooxyethylene units capable of forming 14-, 17and 20-membered rings, resp. Although the effect of alkylaluminum chlorides was gradually reduced with increasing ring size, it was remarkable in the formation of 14-membered rings for methacrylates and even the 20-membered rings for acrylates. When combined with previous results for the analogs containing oligomethylene units, the present data showed that oxyethylene groups had favorable effects on cyclopolymn. A plot of the extent of cyclization against ring size showed that monomers containing oligooxyethylene units had a greater cyclization tendency than those containing oligomethylene units. Plotting the log of the rate consts. for linear propagation and cyclization against ring size gave a straight line in the ring-size range from 11 to 20 for acrylates.

77504-03-1P 77504-05-3P 77504-07-5P 90833-60-6P 90883-61-7P 90883-62-8P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and properties and cyclopolymn. of)

77504-03-1 HCAPLUS RN

ТТ

CN 2-Propenoic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

77504-05-3 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 77504-07-5 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

PAGE 1-B

=CH2

RN 90883-60-6 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-[2-(2propenyl)phenoxy]ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 90883-61-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-[2-[2-(2propenyl)phenoxy]ethoxy]ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 90883-62-8 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-[2-(2-

propenyl)phenoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

$$\begin{array}{c} \text{CH}_2-\text{CH}_$$

CC 35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 67

IT 77504-03-12 77504-05-32 77504-07-52 90883-60-62 90883-61-72 90883-62-82

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);

RACT (Reactant or reagent)

(preparation and properties and cyclopolymn. of)

L35 ANSWER 22 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1984:85273 HCAPLUS Full-text

DOCUMENT NUMBER: 100:85273

ORIGINAL REFERENCE NO.: 100:12917a,12920a

TITLE: Diethylene glycol methacrylate allyl ether.

INVENTOR(S): Voronina, T. A.; Fomina, N. V.

PATENT ASSIGNEE(S): USSR

SOURCE: U.S.S.R. From: Otkrytiya, Izobret., Prom.

Obraztsy, Tovarnye Znaki 1983, (38), 85.

CODEN: URXXAF

DOCUMENT TYPE: Patent LANGUAGE: Russian

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
SU 1047898	A1	19831015	SU 1982-3435735	
				198205
				11
PRIORITY APPLN. INFO.:			SU 1982-3435735	
				198205
				11

Title substance is prepared from diethylene glycol and allyl bromide by treating the resulting monoallyl ether of diethylene glycol with methacrylic chloride in the presence of a catalyst at $0-(-5)^{\circ}$. The procedure is simplified and the yield is increased by condensing diethylene glycol with allyl bromide at $90-140^{\circ}$ in the presence of metallic Cu in a polar solvent consisting of DMF or DMSO with subsequent cooling of the reaction material containing the monoallyl ether of diethylene glycol up to $0-(-5)^{\circ}$ and addition of methacrylic chloride.

IT 58385-94-78

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN 58985-94-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-(2-propenyloxy)ethoxy]ethyl ester (9CI) (CA INDEX NAME)

O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-CH

IC C07C069-587; C07C069-54

CC23-17 (Aliphatic Compounds)

ΙT 58985-94-7P

> RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

L35 ANSWER 23 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 1981:175657 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 94:175657

ORIGINAL REFERENCE NO.: 94:28725a,28728a

TITLE:

Effect of alkylaluminum chlorides on the cyclopolymerization of unconjugated dienes AUTHOR(S): Yokota, Kazuaki; Kakuchi, Toyoji; Takada,

Yoshiyuki

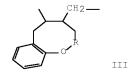
CORPORATE SOURCE: Fac. Eng., Hokkaido Univ., Sapporo, Japan

SOURCE: Hokkaido Daigaku Kogakubu Kenkyu Hokoku (1980),

(102), 45-54

CODEN: HDKKAA; ISSN: 0385-602X

DOCUMENT TYPE: Journal LANGUAGE: Japanese GΙ



AB Unconjugated dienes such as ω -(2-allylphenoxy)alkyl acrylates [o-CH2:CHCH2C6H4O(CH2CH2)nO2CCH:CH2, n = 1,2,3,5] (I), and ω -(2allylphenyl)oligooxyethylene acrylates [o-CH2:CHCH2C6H4(OCH2CH2)nO2CCH:CH2, n = 0,2,3,4] (II) were polymerized in the presence of AlEt2Cl [96-10-6], AlEt1.5Cl1.5 [12075-68-2], and AlEtCl2 [563-43-9] to give polymers containing 7-20 membered rings [III, R = O(CH2CH2)n or (OCH2CH2)n]. The catalysts increased the reaction rate and the extent of cyclization in the case of I (n = 1, 2, 3) or II (n = 0, 2, 3) but for making larger rings, they were not effective. Copolymn. studies with 4-chlorostyrene [1073-67-2] suggested that the catalysts interact with both the double bonds in the same monomer

77504-03-1 77504-05-3 77504-07-5 ТТ

RL: RCT (Reactant); RACT (Reactant or reagent) (cyclopolymn. of, in presence of ethylaluminum chlorides, mechanism of)

77504-03-1 HCAPLUS RN

2-Propenoic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 77504-05-3 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 77504-07-5 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-[2-(2-propenyl)phenoxy]ethoxy]et hoxy]ethyl ester (9CI) (CA INDEX NAME)

PAGE 1-B

—CH2

IT 77504-04-2P 77504-06-4P 77504-08-6P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN 77504-04-2 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 77504-03-1

CMF C16 H20 O4

RN 77504-06-4 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 77504-05-3 CMF C18 H24 O5

RN 77504-08-6 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 77504-07-5 CMF C20 H28 O6

PAGE 1-B

—CH2

CC 35-4 (Synthetic High Polymers)

IT 14925-75-8 61632-59-5 61632-60-8 77504-03-1 77504-05-3 77504-07-5 77504-09-7 77505-38-5

RL: RCT (Reactant); RACT (Reactant or reagent)

(cyclopolymn. of, in presence of ethylaluminum chlorides,

mechanism of)
IT 27101-85-5P 77504-04-2P 77504-06-4P

3/11/2008 10/588,210 77504-08-6P 77504-10-0P 77504-11-1P 77504-12-2P 77538-73-9P RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of) L35 ANSWER 24 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 1979:169022 HCAPLUS Full-text ACCESSION NUMBER: DOCUMENT NUMBER: 90:169022 ORIGINAL REFERENCE NO.: 90:26851a,26854a Synthesis of esters of acrylic and methacrylic TITLE: acids containing allyl groups AUTHOR(S): Beshenova, E. P.; Etlis, V. S. CORPORATE SOURCE: USSR SOURCE: Osnovnoi Organicheskii Sintez i Neftekhimiya (1977), 8, $\overline{5}7-9$ CODEN: OOSNDC; ISSN: 0321-2386 DOCUMENT TYPE: Journal LANGUAGE: Russian CASREACT 90:169022 OTHER SOURCE(S): The syntheses are described of 8 title compds. CH2:CRCO2ZCH2CH:CH2 [R = H, Me; Z = CH2CH2OCO2CH2CH2O, CH2CH(OH)CH2O, CH2CH(CH2OCH2CH:CH2)OCO2, CH(CH2OCH2CH:CH2)CH2O] from acrylic and methacrylic acids, their derivs., and derivs. of allyl alc. 69936-66-9P 69936-67-0P RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of) RN 69936-66-9 HCAPLUS 2-Propenoic acid, 2-[[[2-(2-propenyloxy)ethoxy]carbonyl]oxy]ethyl ester (9CI) (CA INDEX NAME) -сн2-сн2-RN 69936-67-0 HCAPLUS 2-Propenoic acid, 2-methyl-, 2-[[[2-(2-propenyloxy)ethoxy]carbonyl]o xy]ethyl ester (9CI) (CA INDEX NAME) CH2-CH2-O-CH2-CH=-CH2 35-2 (Synthetic High Polymers) Section cross-reference(s): 23 22214-16-0P 22214-17-1P 69936-66-9P 69936-67-0P

69936-68-1P 69936-69-2P RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of) L35 ANSWER 25 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 1976:151442 HCAPLUS Full-text ACCESSION NUMBER: DOCUMENT NUMBER: 84:151442 ORIGINAL REFERENCE NO.: 84:24621a,24624a TITLE: Crosslinked hydrogel copolymer material INVENTOR(S): Howes, John G. B.; Da Costa, Nicholas M.; Selway, Rupert A.; Potter, William D.

10/588,210

PATENT ASSIGNEE(S): SOURCE:

Smith and Nephew Research Ltd., UK

55

Deleted: 3/6/2008

Ger. Offen., 17 pp. CODEN: GWXXBX

DOCUMENT TYPE: LANGUAGE: FAMILY ACC. NUM. COUNT: 1

Patent German

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 2529639	A1	19760122	DE 1975-2529639	197507 03
DE 2529639 GB 1494641	C2 A	19850725 19771207	GB 1974-29757	197407
ZA 7504036	A	19760526	ZA 1975-4036	197506
US 4036814	A	19770719	US 1975-590037	24 197506 25
NO 7502351	A	19760106	NO 1975-2351	197506 30
NO 144888 NO 144888 DK 7503006	B C A	19810824 19811202 19760105	DK 1975-3006	197507
NL 7507914	A	19760106	NL 1975-7914	197507
FR 2277110	A1	19760130	FR 1975-20881	03 197507 03
FR 2277110 AU 7582722	B1 A	19801003 19770106	AU 1975-82722	197507 03
CH 603708	A5	19780831	CH 1975-8688	197507 03
CS 194231	В2	19791130	CS 1975-4749	197507 03
BE 831047	A1	19760105	BE 1975-158027	197507 04
FI 7501961	A	19760105	FI 1975-1961	197507 04
FI 59114 FI 59114 SE 7507693	B C A	19810227 19810610 19760105	SE 1975-7693	197507
SE 407416	С	19790705		04

3/11/2008	10/588,210			56	Deleted: 3/6/2008
SE 407416	3 19790326				
JP 51030750 A	A 19760316	JP 1975-82013			
			197507		
			04		
	3 19831012				
BR 7504239	A 19760706	BR 19 7 5-5 4 29			
			197507		
			0.4		
DD 123396	A5 19761212	DD 1975-187112			
			197507		
			0.4		
CA 1037196	A1 19780822	CA 1975-230799			
			197507		
			04		
PRIORITY APPLN. INFO.:		GB 1974-29757	A		
INTOTALLI PALLEN. INTO		GB 1371 23737	197407		
			04		
			04		
		GB 1975-17586	A		
		GB 1973-17300			
			197504		
			28		

AB Phenylethyl methacrylate (I), benzyl methacrylate, phenoxyethyl methacrylate, β -naphthyl methacrylate, or a similar methacrylate was copolymd. with vinylpyrrolidone (II) and with allyl methacrylate (III), 3-allyloxy-2-hydroxypropyl methacrylate, 2-allyloxyethyl methacrylate, or a similar crosslinking monomer to prepare crosslinked hydrogel copolymers which absorbed 65-85% of a physiol. salt solution and were especially useful for the manufacture of contact lenses. Thus, a mixture of I 6.4, II 33.6, III 0.27, and AIBN 0.12 g were heated 24 hr at 45-55° and 1-10 hr at 110° to prepare a copolymer [58986-09-7].

IT 58985-95-8 58985-97-0 58985-98-1 58985-99-2 58986-00-8 58986-01-9 58986-03-1 58986-04-2 58986-05-3 58986-08-6

RL: USES (Uses) (hydrogels)

RN 58985-95-8 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-phenylethyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-(2-propenyloxy)ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-94-7 CMF C11 H18 O4

CM 2

CRN 3683-12-3 CMF C12 H14 O2

CM3

CRN 88-12-0 CMF C6 H9 N O



58985-97-0 HCAPLUS

RN2-Propenoic acid, 2-methyl-, 2-phenylethyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM1

CRN 58985-96-9 CMF C13 H22 O5

CM2

CRN 3683-12-3 CMF C12 H14 O2

CM3

CRN 88-12-0 CMF C6 H9 N O



RN 58985-98-1 HCAPLUS
CN 2-Propenoic acid, 2-methyl-, phenylmethyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-96-9 CMF C13 H22 O5

CM 2

CRN 2495-37-6 CMF C11 H12 O2

CM 3

CRN 88-12-0 CMF C6 H9 N O

58985-99-2 HCAPLUS

2-Propenoic acid, 2-methyl-, phenyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-96-9

CMF C13 H22 O5

$$\begin{array}{c} {\rm H2C} \\ {\rm C} \\ {\rm C$$

CM 2

CRN 2177-70-0 CMF C10 H10 O2

CM 3

CRN 88-12-0 CMF C6 H9 N O



RN 58986-00-8 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-phenoxyethyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-96-9 CMF C13 H22 O5

$$\begin{array}{c} {}_{\rm H\,2\,C} \circ \\ {}_{\rm M\,e} - {}_{\rm C} - {}_{\rm C} \circ - {}_{\rm C\,H\,2} - {}_{$$

CM 2

CRN 10595-06-9 CMF C12 H14 O3

CM 3

CRN 88-12-0 CMF C6 H9 N O

RN 58986-01-9 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 4-methoxyphenyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-96-9 CMF C13 H22 O5

CM 2

CRN 10430-85-0 CMF C11 H12 O3

$$\begin{array}{c} \circ \quad \overset{\circ}{\mathbb{C}} \quad \overset{\overset}{\mathbb{C}} \quad \overset{\overset}{\mathbb{$$

CM 3

CRN 88-12-0 CMF C6 H9 N O

RN 58986-03-1 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, (4-methoxyphenyl)methyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58986-02-0 CMF C12 H14 O3

CM 2

CRN 58985-96-9 CMF C13 H22 O5

$$\begin{array}{c} ^{\rm H2C} \circ \\ \\ ^{\rm Me} - ^{\rm C} - ^{\rm C} \circ - ^{\rm CH}_2 - ^$$

CM 3

CRN 88-12-0 CMF C6 H9 N O

RN 58986-04-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-naphthalenyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

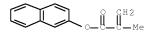
CM 1

CRN 58985-96-9 CMF C13 H22 O5

$$\begin{array}{c} ^{\rm H2C} \circ \\ ^{\rm Me} - \overset{\circ}{U} - \overset{\circ}{U} - \circ - \circ + 2 - \circ + 2 - \circ - \circ + 2 - \circ + 2 - \circ - \circ + 2 -$$

CM 2

CRN 10475-46-4 CMF C14 H12 O2



CM 3

CRN 88-12-0 CMF C6 H9 N O



RN 58986-05-3 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, [1,1'-biphenyl]-4-yl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-96-9 CMF C13 H22 O5

CM 2

CRN 46904-74-9 CMF C16 H14 O2

CM 3

CRN 88-12-0 CMF C6 H9 N O

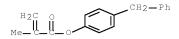


RN 58986-07-5 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 4-(phenylmethyl)phenyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58986-06-4 CMF C17 H16 O2



CM 2

CRN 58985-96-9 CMF C13 H22 O5

$$\begin{array}{c} ^{\rm H2C} \circ \\ ^{\rm Me} - \overset{\circ}{U} - \overset{\circ}{C} - \circ - \circ + 2 - \circ + 2 - \circ - \circ + 2 - \circ + 2 - \circ - \circ + 2 -$$

CM 3

CRN 88-12-0 CMF C6 H9 N O



RN 58986-08-6 HCAPLUS

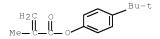
CN 2-Propenoic acid, 2-methyl-, 4-(1,1-dimethylethyl)phenyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-96-9 CMF C13 H22 O5

CM 2

CRN 13101-33-2 CMF C14 H18 O2



CM 3

CRN 88-12-0 CMF C6 H9 N O



IC C08F; G02C

CC 36-3 (Plastics Manufacture and Processing)

Section cross-reference(s): 63

IT 58985-90-3 58985-91-4 58985-92-5 58985-93-6 58985-95-8

58985-97-0 58985-98-1 58985-99-2

58986-00-8 58986-01-9 58986-03-1

58986-04-2 58986-05-3 58986-07-5

58386-03-6 58986-09-7

RL: USES (Uses)
(hydrogels)

L35 ANSWER 26 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1955:27989 HCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 49:27989
ORIGINAL REFERENCE NO.: 49:5388e-i

TITLE: Three dimensional polymerization of allyl ethers and mixed allyl ethers of methacrylic esters of

glycols

AUTHOR(S): Berlin, A. A.; Dabagova, A. K.; Rodionova, E. F. SOURCE: Sbornik Statei Obshchei Khim. (1953), 2, 1560-5

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

Reaction of glycols with CH2:CMeCOCl or CH2:CHCH202CCl in the presence of pyridine at -10° gave 60-70% of the following derivs.: CH2:CHCH2OCH2CH2O2CCMe:CH2, b2 69-70°, d2020 0.9706, nD20 1.4460; CH2:CHCH2OCH2CH2OCH2CH2O2CCMe:CH2, b2 85-6°, d2020 1.0270, nD20 1.4538; CH2:CHCH2O(CH2CH2O)2CH2CH2O2CCMe:CH2, b. undetd. owing to polymerization, d2020 1.0550, nD20 1.4587; CH2:CMeCO2CH2CH2CD2CCH2CH:CH2, b2 106°, d2020 1.0820, nD20 1.4500; CH2; CMeCO2(CH2)2O(CH2)2CO2CH2CH: CH2 b2 119-20°, d2020 1.1020, nD20 1.4560; CH2:CMeCO2(CH2CH2O)3COCH2CH:CH2, polymerizes on attempted distillation, d2020 1.1070, nD20 1.4585; (CH202CCH2CH:CH2)2, b2 127.5°, d2020 1.1210, nD20 1.4445; O(CH2CH2O2CCH2CH:CH2)2, b2 161°, d2020 1.1400, nD20 1.4515; CH2:CHCH2OCH2CH2CH2CH2CH2CH2CH2, b2 85.5°, d2020 1.0384, nD20 1.4415. The intermediate allyl ethers were prepared from RCl or RBr and the corresponding Na derivative of the glycols; HOCH2CH2CCH2CH2CH2 (I), b. 159-60°, d2020 0.9526, nD20 1.4355; HO(CH2CH2O)2CH2CH:CH2, b2 98-101°, d2020 1.012, nD20 1.4440; HO(CH2CH2O)3CH2CH:CH2, b2 115-18°, d2020 1.0699, nD20 1.4530. Passage of ethylene oxide into CH2:CHCH2OH and 3% concentrated H2SO4 at $50-60\,^{\circ}$ gave 50-5% yield of I. Polymerization of these esters were run in pure state and in 25% MeOH solns. The results, given graphically, show the following. The methacrylicallyl derivs. of the glycols and methacrylic-"carballylic" derivs. polymerize more rapidly than do "biscarballylic" or allyl "carballylic" derivs. Generally the increase of the distance between the functional groups of the above esters leads to increase rate of 3-dimensional polymerization; in "biscarballylic" esters this relationship is reversed. The principal factor affecting the rate of polymerization in MeOH is the steric factor which establishes the distance between the functional groups of the monomer.

IT 58985-94-7, Ethanol, 2-[2-(allyloxy)ethoxy]-, methacrylate 58985-96-9, Ethanol, 2-[2-[2-(allyloxy)ethoxy]ethoxy]-, methacrylate

(polymerization of) RN 58985-94-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-(2-propenyloxy)ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 58985-96-9 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-[2-(2-propenyloxy)ethoxy]ethyl ester (9CI) (CA INDEX NAME)

CC 10 (Organic Chemistry)

IT 15075-50-0, Ethanol, 2-[2-(allyloxy)ethoxy] - 26150-05-0, Ethanol, 2-[2-[2-(allyloxy)ethoxy] - 44605-74-5, Carbonic acid, allyl ester 58985-94-7, Ethanol, 2-[2-(allyloxy)ethoxy] -, methacrylate 58985-96-9, Ethanol, 2-[2-[2-(allyloxy)ethoxy] -, (allyloxy)ethoxy] -, methacrylate 58985-96-9, 3,6,9-Trioxadodec-11-en-1-ol, methacrylate (polymerization of)

=>